

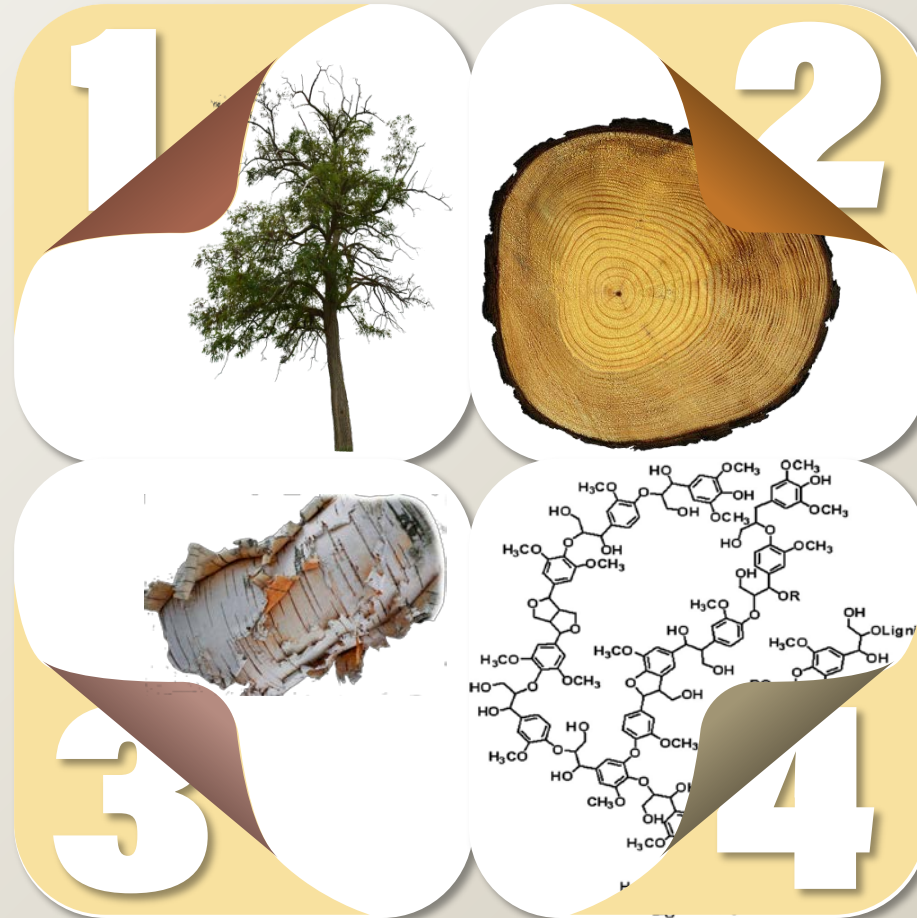
# Selected wood chemical properties – bridge to technology

Radko Tiňo



01

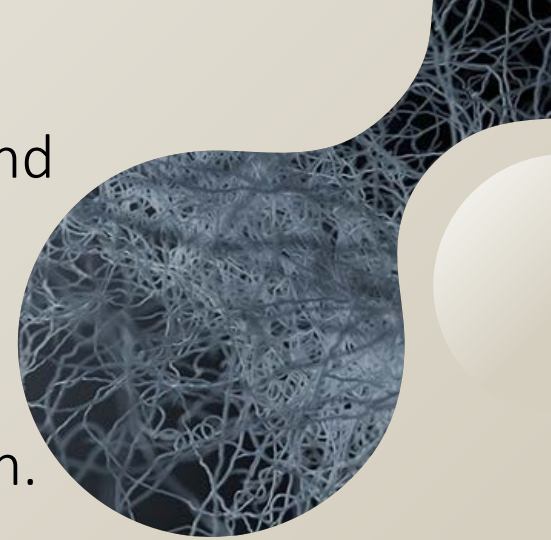
# What is the wood?

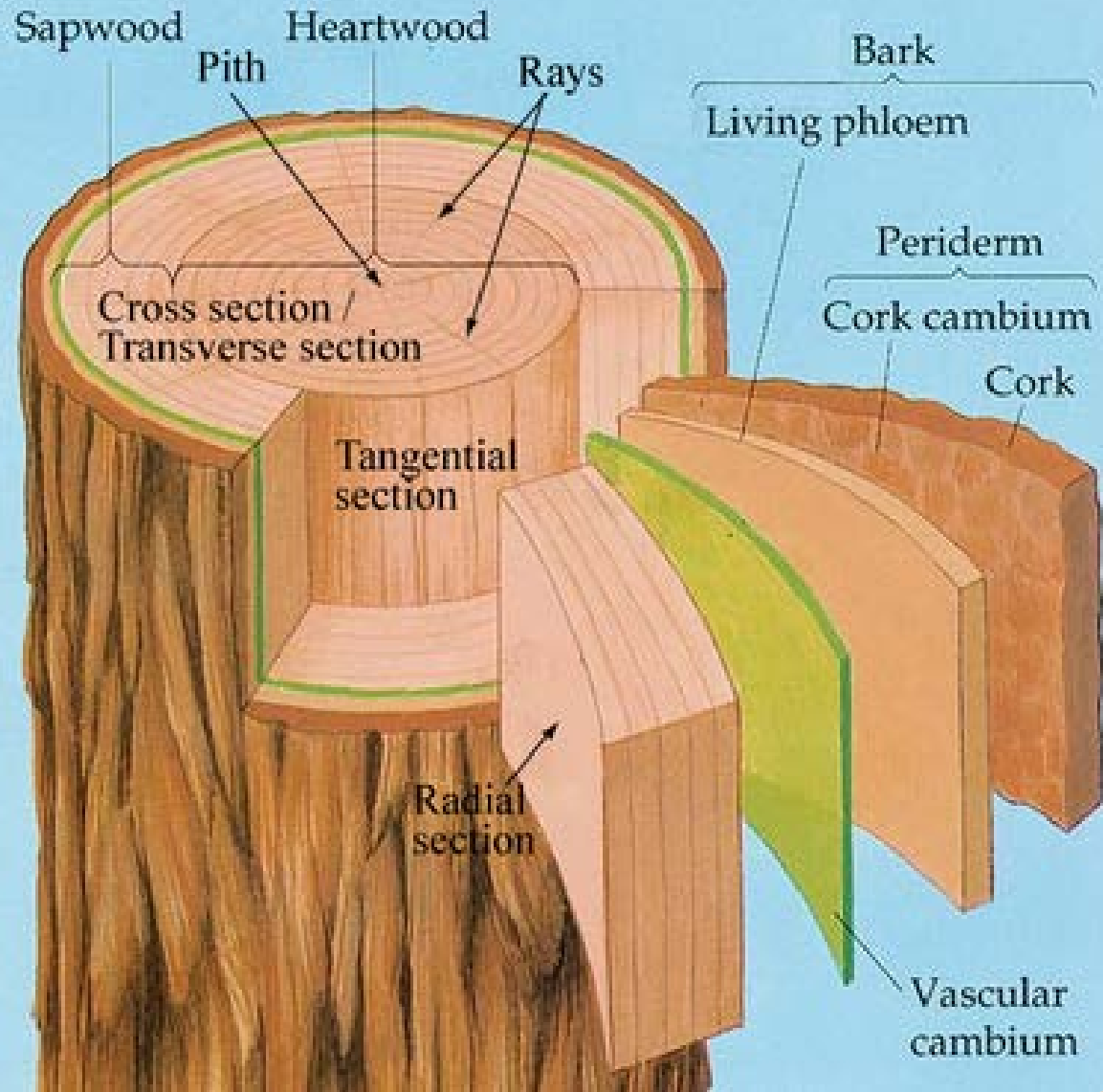


Wood is **complex** porous and fibrous structural **tissue** found in the stems and roots of trees and other **woody** plants.

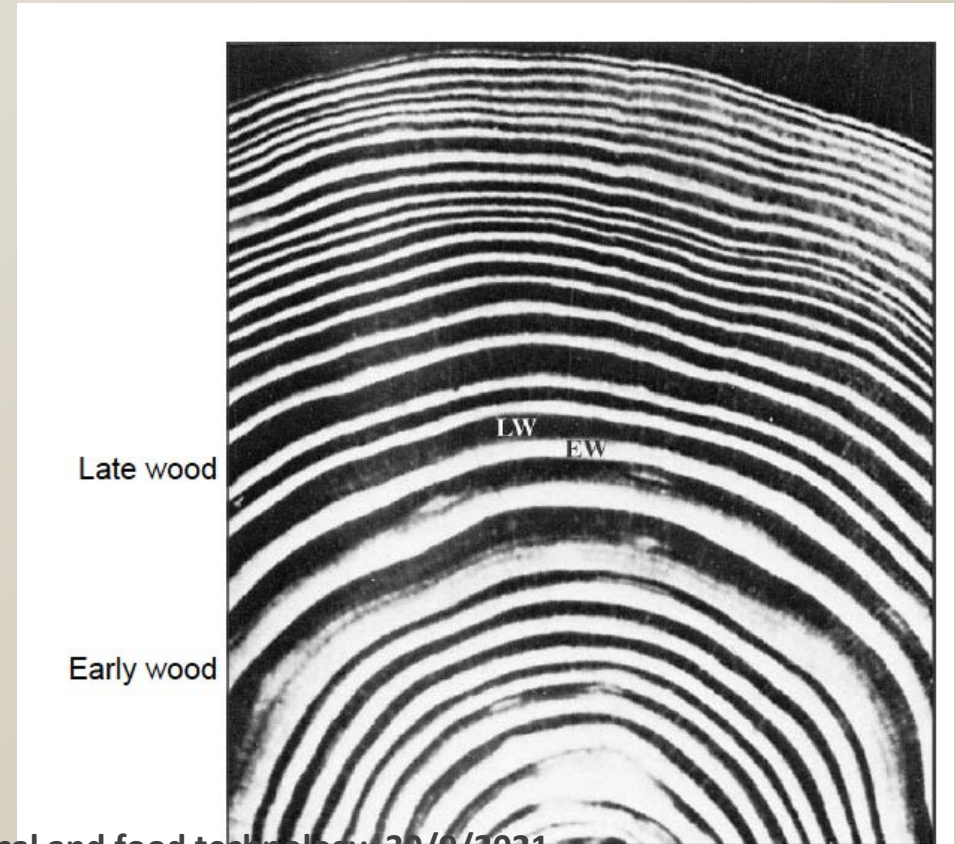
It is an organic material, a **natural composite of cellulose fibers** (which are strong in tension) embedded in a **matrix of lignin** which resists compression.

Wood has **extreme anisotropy** because **90 to 95%** of all the cells are **elongated and vertical** (i.e. *aligned parallel to the tree trunk*). The remaining 5 to 10% of cells are arranged in **radial directions**, with **no cells** at all aligned tangentially.

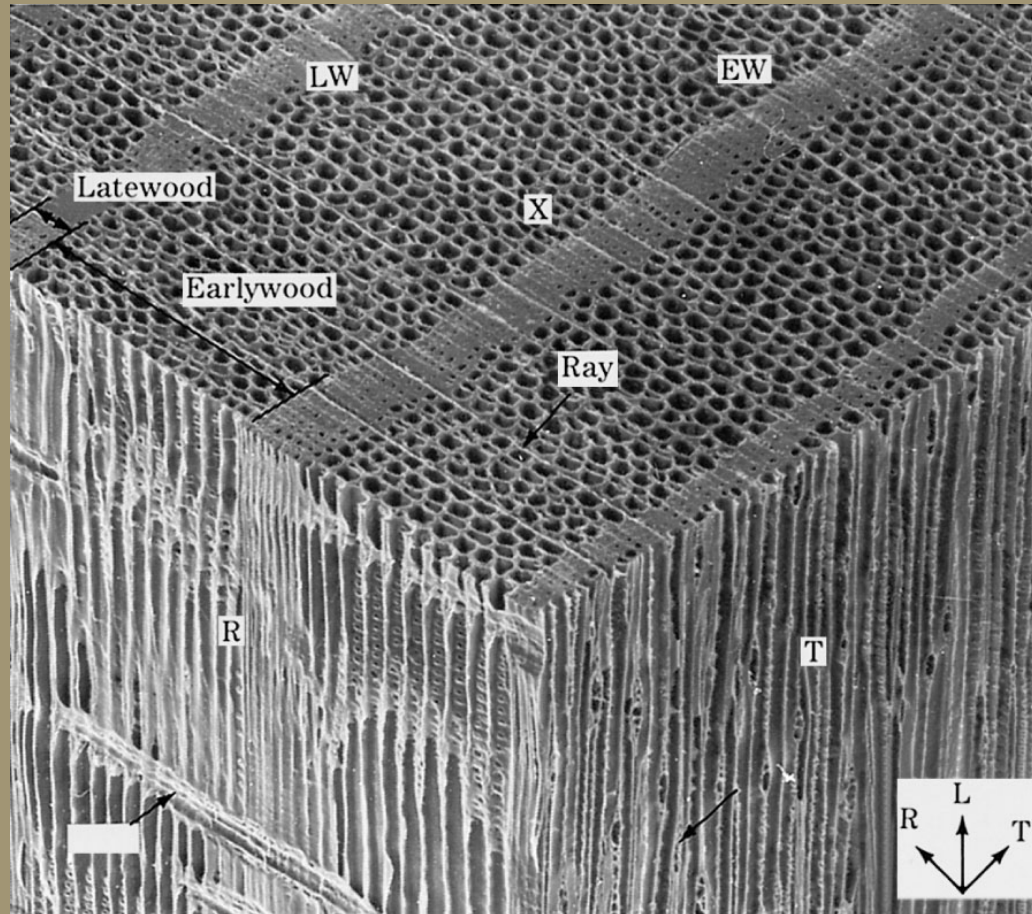




**Three main sections**  
**the heartwood**, which is *physiologically inactive*,  
**the sapwood**, where all conduction and storage occurs,  
**the bark**, which *protects* the interior of the tree trunk.

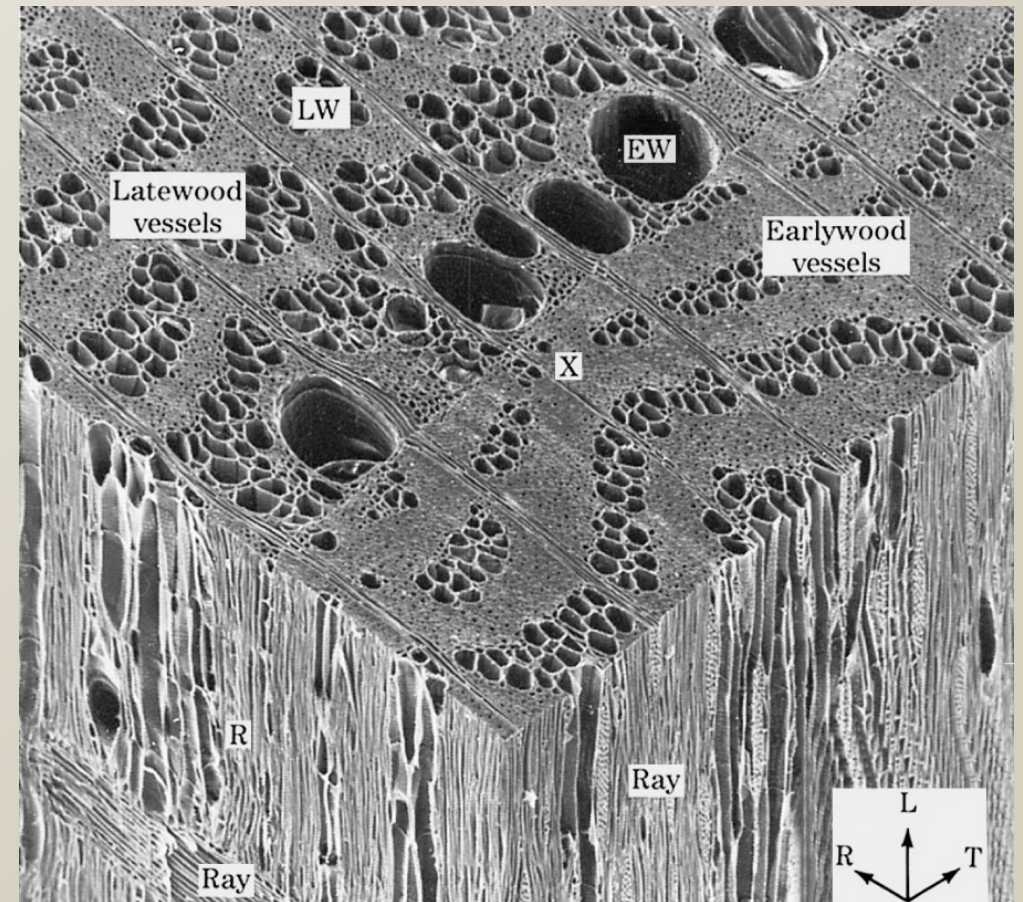
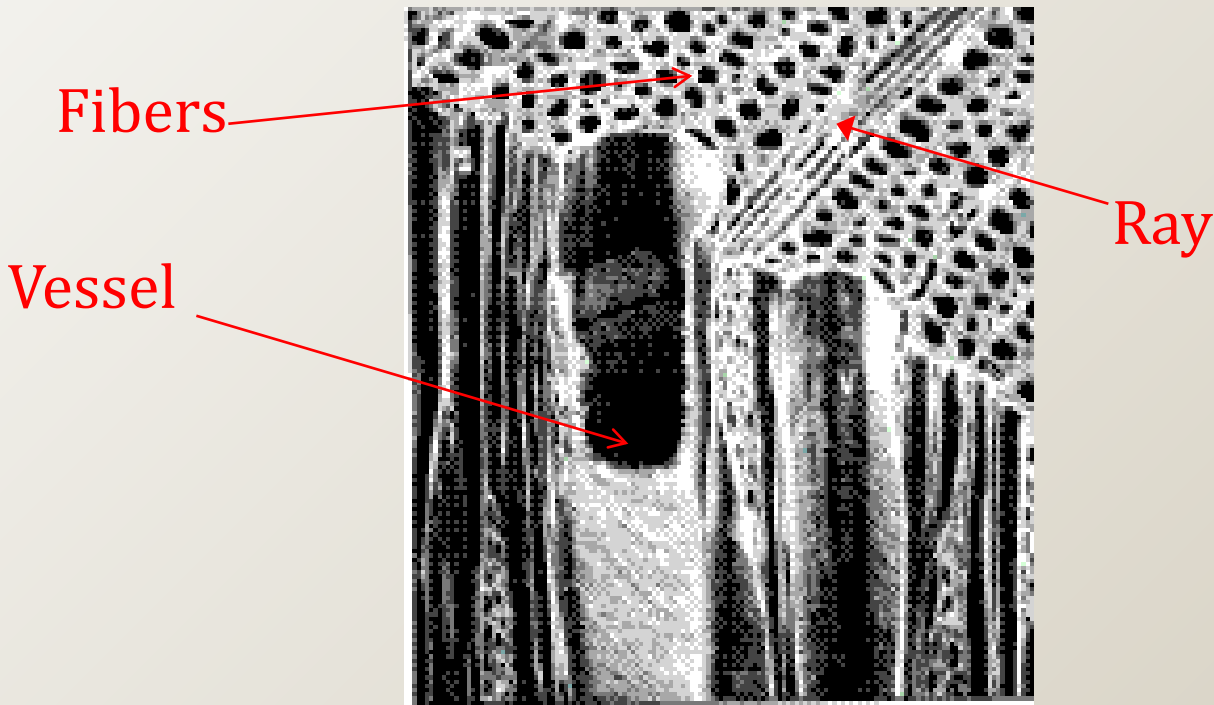


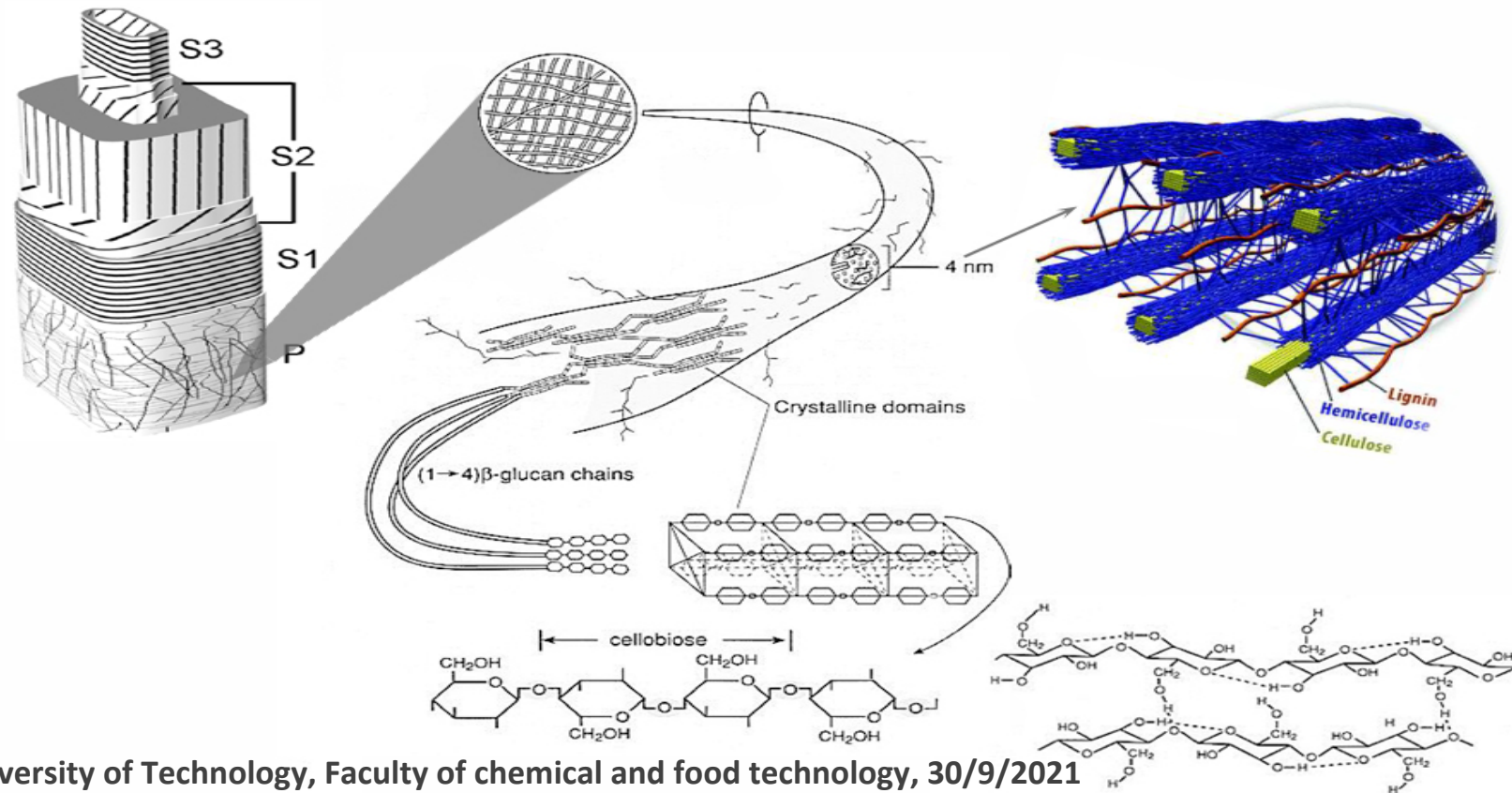
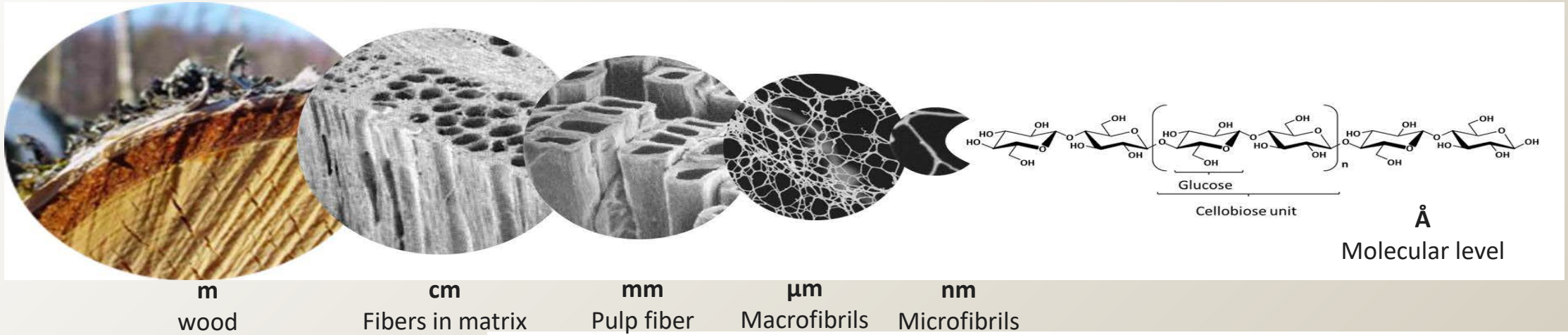
Softwoods consist mainly of long (3 to 5 mm) cells called tracheids which are about 20 to 80 x 10<sup>-6</sup>m.



**Hardwoods** consist mainly of two kinds of cells:

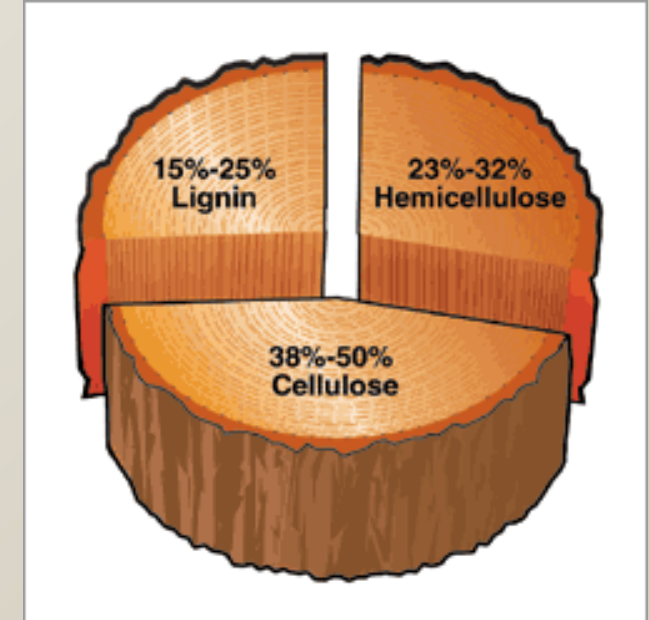
- **Wood fibers** are elongated cells which are similar to tracheids except they are smaller, only 0.7 to 3 mm long and less than  $20 \times 10^{-6} \text{m}$  in diameter, and they do not serve for fluid transport in the living tree.
- **The vessel elements** do serve for fluid transport in the living tree, and they can have a wide range of sizes.





# Classification of substances in wood

- Main components (90 -97%)
  - Saccharidic (70%)
    - Cellulose
    - hemicelluloses
  - Aromatic (25%) - lignin
- Accessoric components (3-10%) (extractives)
  - Inorganic (ash) (0,5%)
  - Organic – saccharides. Phenols, terpenes, acyclic acids, alcohols, proteins, etc.





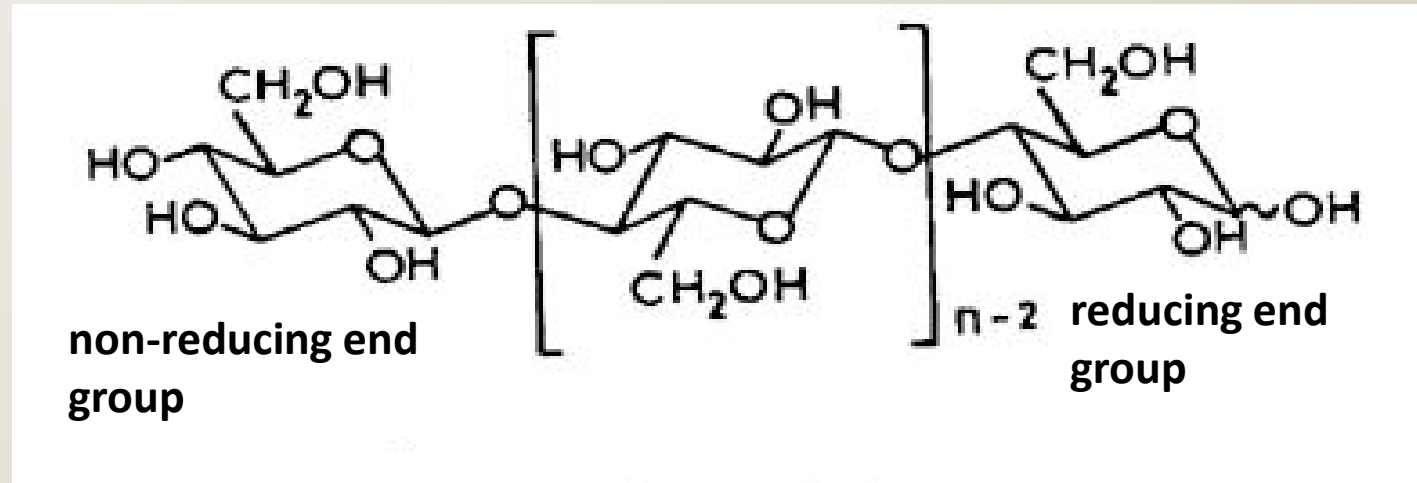
# Chemical Constituents of Wood

The chemical composition of wood varies from species to species, but is approximately 50% carbon, 42% oxygen, 6% hydrogen, 1% nitrogen, and 1% other elements (mainly *calcium*, *potassium*, *sodium*, *magnesium*, *iron*, and *manganese*) by weight.

Wood also contains *sulfur*, *chlorine*, *silicon*, *phosphorus*, and other elements in small quantity.

# Chemical Constituents of Wood

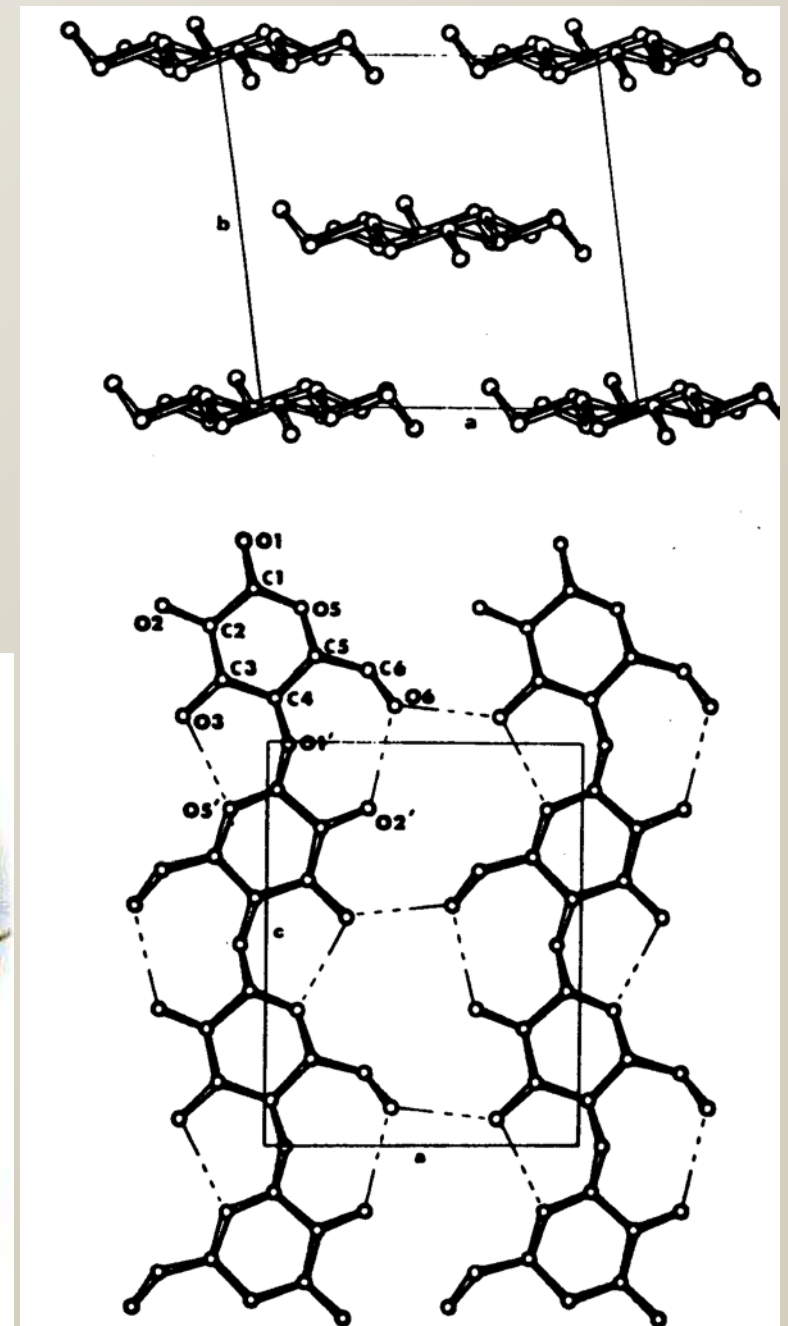
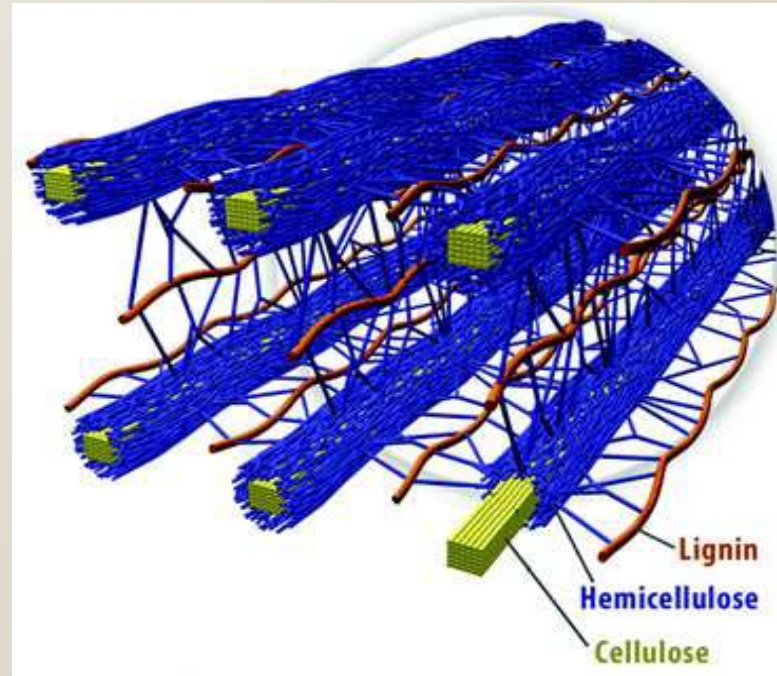
## Cellulose



- linear homopolysaccharide
- Cellulose is a glucan polymer consisting of linear chains of 1,4- $\beta$ -bonded anhydroglucose units. *(The notation 1,4- $\beta$  describes the bond linkage and the configuration of the oxygen atom between adjacent glucose units.)*
- The number of sugar units in one molecular chain is referred to as the degree of polymerization (DP).
- A DP of 10,000 would mean a linear chain length of approximately 5  $\mu\text{m}$  in wood.
- Cellulose in the primary wall has a lower DP and is thought to be polydisperse.

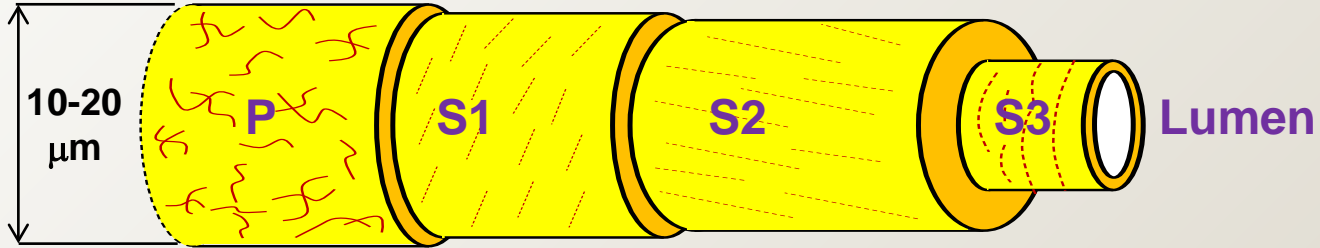
- Native cellulose is partially crystalline.
- The unit cell contains eight cellobiose moieties. The molecular chains pack in layers that are held together by weak van der Waals' forces
- The layers consist of parallel chains of **anhydroglucose units**, and the chains are held together by **intermolecular hydrogen bonds**. There are also intramolecular hydrogen bonds between the atoms of adjacent glucose residue
- This structure is called cellulose I.

- Cellulose is **insoluble in most solvents** including strong alkali.
- It is **difficult to isolate from wood in pure form** because it is intimately associated with the **lignin and hemicelluloses**.

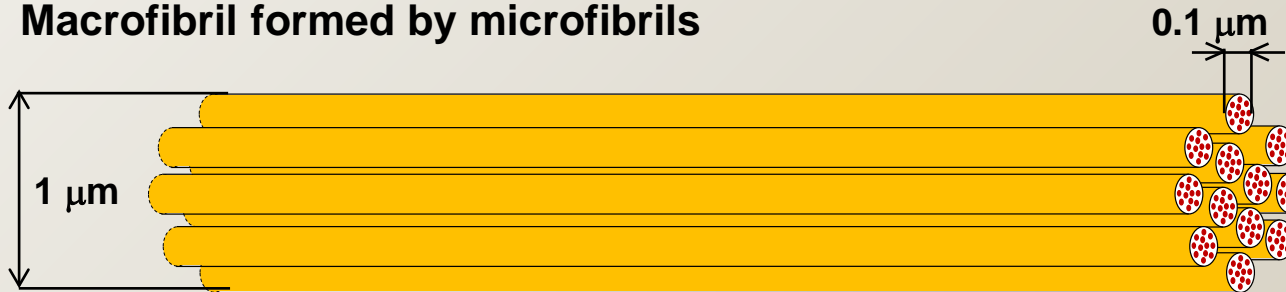


# Dimensions of the hierarchical cellulose structure

Fiber wall sublayers: **P** – Primary, **S1, S2, S3** – secondary



Macrofibril formed by microfibrils



Bundle of Cellulose chains

10-25 nm

amorphous  
~ 10 nm

crystalline  
~ 20 nm

The lowest chemical stability, degradation



Internal transport: reactants have to reach surface of cellulose chains (nanometric scale)

# Chemical Constituents of Wood

**Hemicelluloses** constitutes about **20% in deciduous trees** but near **30% in conifers**. It is mainly five and six-carbon sugars that are linked in an irregular manner, in contrast to the cellulose.

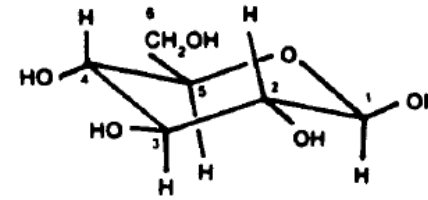
**Lignin** is the third component at around **27% in coniferous** wood vs. **23% in deciduous trees**. Lignin confers the **hydrophobic properties** reflecting the fact that it is based on aromatic rings.

Aside from the „lignocellulose“, wood consists of a variety of **low molecular weight organic compounds**, called **extractives**.

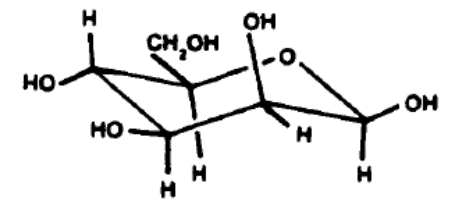
The wood extractives are **fatty acids, resin acids, waxes** and **terpenes**. For example, **resin** is exuded by **conifers** as **protection from insects**.

# Hemicelluloses

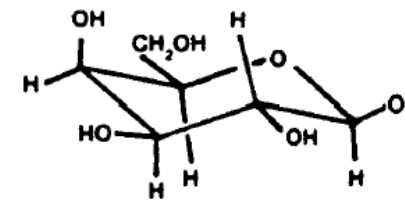
- linear polysaccharides partially branched, containing short side chains
- mixtures of polysaccharides synthesized in wood almost entirely from glucose, mannose, galactose, xylose, arabinose, 4-*O* methylglucuronic acid, and galacturonic acid residues.
- accompanying cellulose in layers of the cell wall
- They are intimately associated with cellulose and appear to contribute as a structural component in the plant.
- Some hemicelluloses are present in abnormally large amounts when the plant is under stress; e.g., compression wood has a higher than normal galactose content as well as a higher lignin content.
- Hemicelluloses are soluble in alkali and easily hydrolyzed by acids.



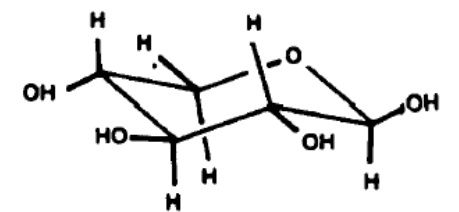
$\beta$ -D-Glucose  
 $\beta$ -D-Glucopyranose  
 $\beta$ -D-Glup



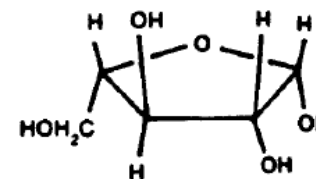
$\beta$ -D-Mannose  
 $\beta$ -D-Mannopyranose  
 $\beta$ -D-Manp



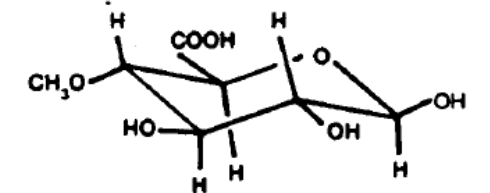
$\beta$ -D-Galactose  
 $\beta$ -D-Galactopyranose  
 $\beta$ -D-Galp



$\beta$ -D-Xylose  
 $\beta$ -D-Xylopyranose  
 $\beta$ -D-Xylp



$\alpha$ -L-Arabinose  
 $\alpha$ -L-Arabinofuranose  
 $\alpha$ -L-Araf



4-*O*-Methylglucuronic acid  
4-*O*-Methylglucopyranosyluronic acid  
4-*O*-Me- $\alpha$ -D-GlupA

<i>Hemicellulose Type</i>	<i>Occurrence</i>	<i>Amount (% of wood)</i>	<i>Composition</i>				<i>Solubility<sup>a</sup></i>	$\overline{DP}_n^b$
			<i>Units</i>	<i>Molar Ratios</i>	<i>Linkage</i>			
Galactoglucomannan	Softwood	5-8	$\beta$ -D-Manp $\beta$ -D-Glup $\alpha$ -D-Galp Acetyl	3 1 1	1 $\rightarrow$ 4 1 $\rightarrow$ 4 1 $\rightarrow$ 6	Alkali, water*	100	
(Galacto)Glucomannan	Softwood	10-15	$\beta$ -D-Manp $\beta$ -D-Glup $\alpha$ -D-Galp Acetyl	4 1 0.1 1	1 $\rightarrow$ 4 1 $\rightarrow$ 4 1 $\rightarrow$ 6	Alkaline borate	100	
Arabinoglucuro- noxylan	Softwood	7-10	$\beta$ -D-Xylp 4-O-Me- $\alpha$ -D- GlupA $\alpha$ -L-Araf	10 2 1.3	1 $\rightarrow$ 4 1 $\rightarrow$ 2 1 $\rightarrow$ 3	Alkali, dimethyl sulfoxide,* water*	100	
Arabinogalactan	Larch wood	5-35	$\beta$ -D-Galp  $\alpha$ -L-Araf $\beta$ -L-Arap $\beta$ -D-GlupA	6  2/3 1/3 Little	1 $\rightarrow$ 3, 1 $\rightarrow$ 6 1 $\rightarrow$ 6 1 $\rightarrow$ 3 1 $\rightarrow$ 6	Water	200	
Glucuronoxylan	Hardwood	15-30	$\beta$ -D-Xylp 4-O-Me- $\alpha$ -D- GlupA Acetyl	10 1 7	1 $\rightarrow$ 4 1 $\rightarrow$ 2	Alkali, dimethyl sulfoxide*	200	
Glucomannan	Hardwood	2-5	$\beta$ -D-Manp $\beta$ -D-Glup	1-2	1 $\rightarrow$ 4 1 $\rightarrow$ 4	Alkaline borate	200	

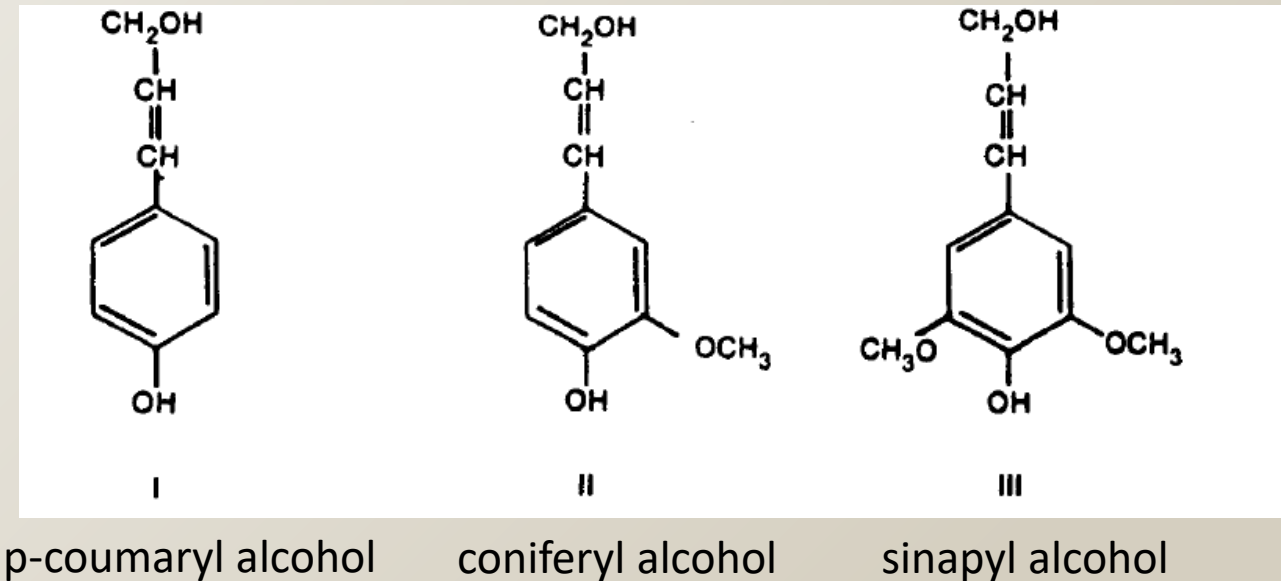
<sup>a</sup> The asterisk represents a partial solubility.

<sup>b</sup>  $\overline{DP}_n$  is the number average degree of polymerization, usually obtained by osmometry.

# Lignin

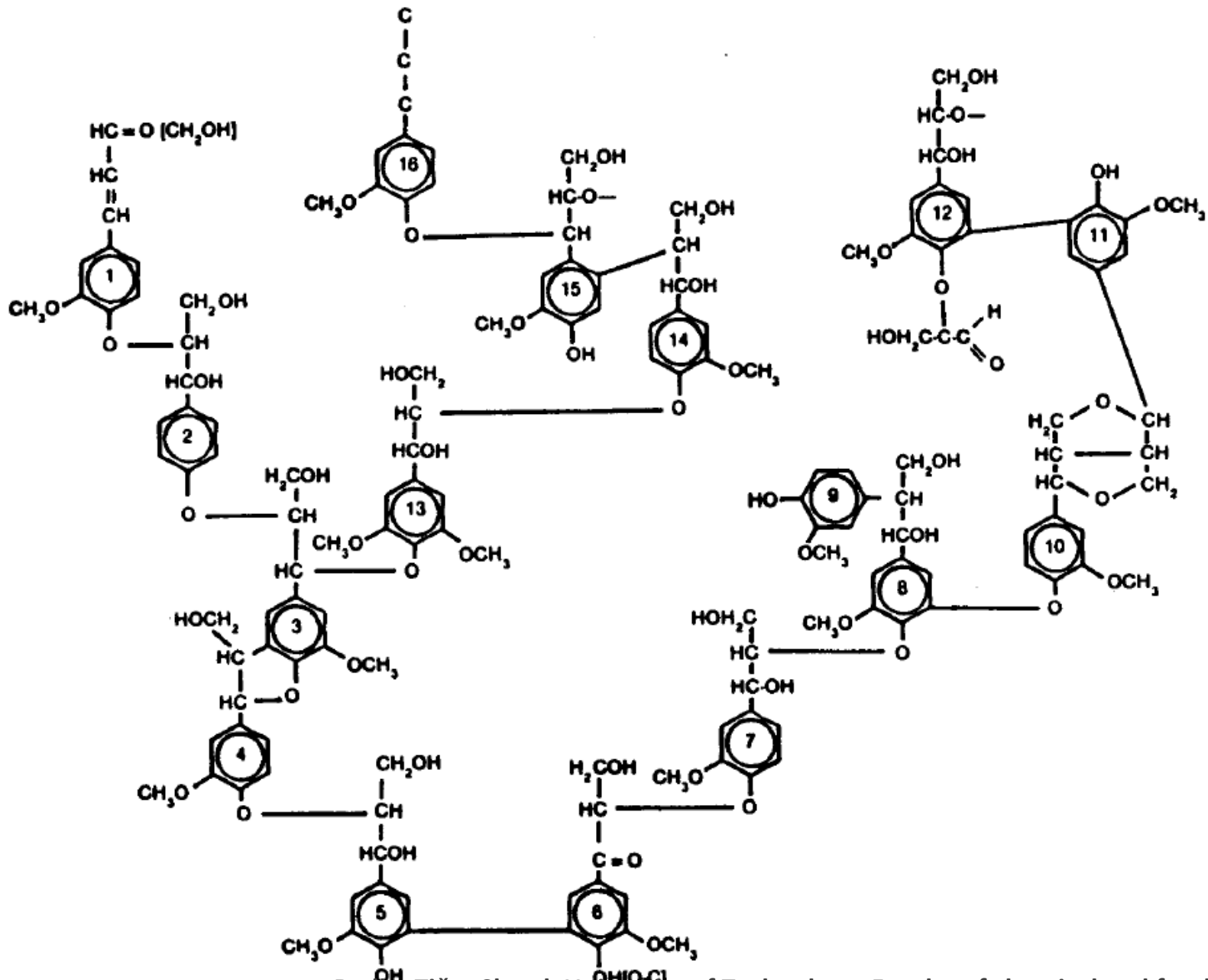
- phenolic substance consisting of an irregular array of variously bonded hydroxy- and methoxy-substituted phenylpropane units.
- I is a minor precursor of softwood and hardwood lignins;
- II is the predominant precursor of softwood lignin;
- II and III are both precursors of hardwood lignin

## The precursors of lignin biosynthesis





## Partial structure of softwood lignin

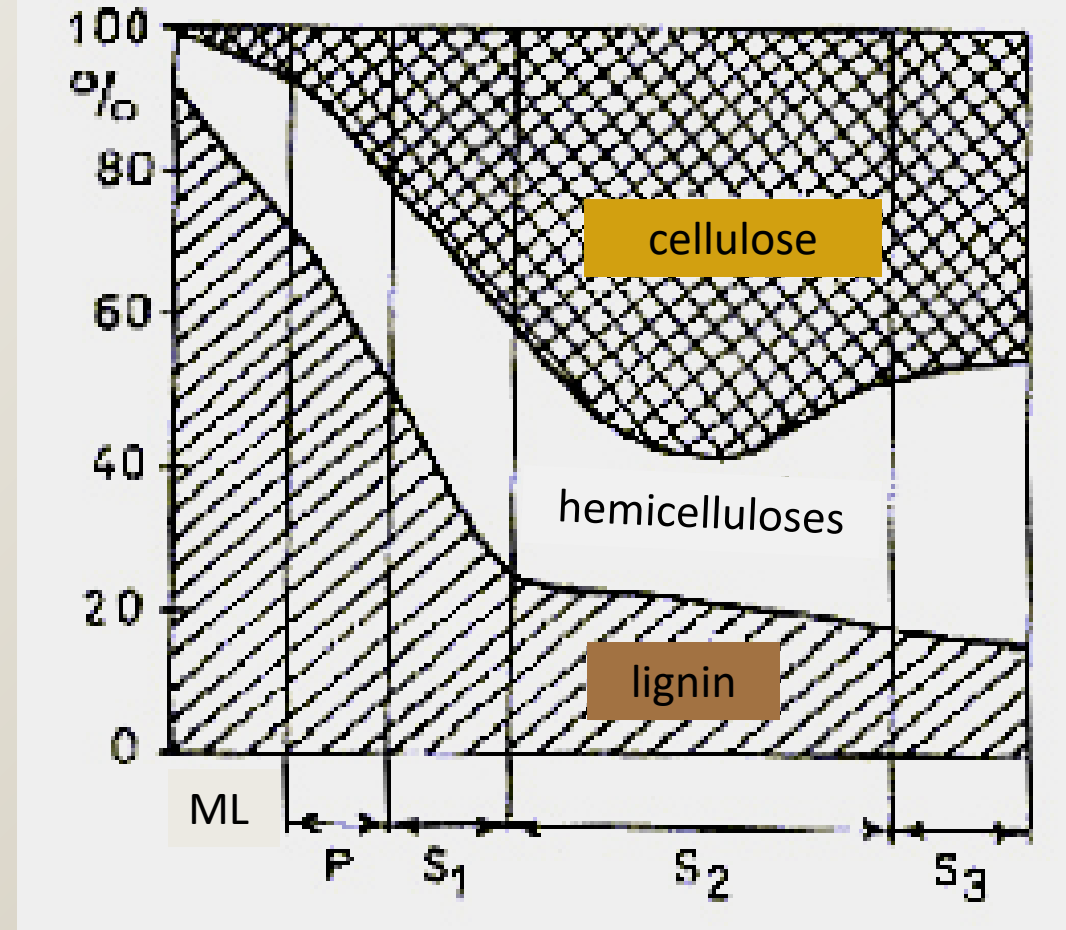
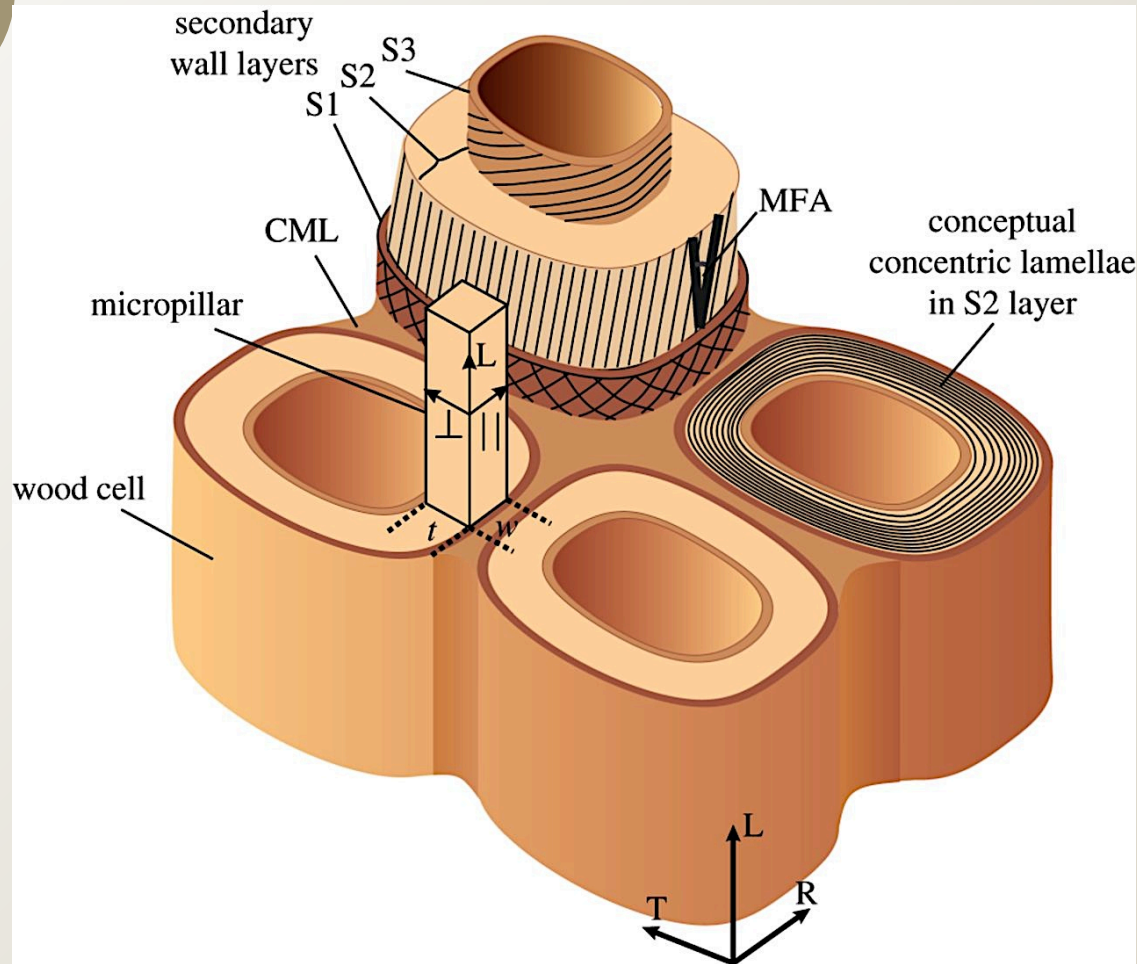


A structure proposed for **hardwood lignin** (*Fagus silvatica* L.) is similar, except that there are three times as many syringylpropane units as guaiacylpropane units

The ratio of syringyl to guaiacyl moieties is often obtained by **measuring the relative amounts** of **syringaldehyde** (3, 5-dimethoxy-4-hydroxybenzaldehyde) and **vanillin** (4-hydroxy-3-methoxybenzaldehyde) generated as products of nitrobenzene oxidation of lignin

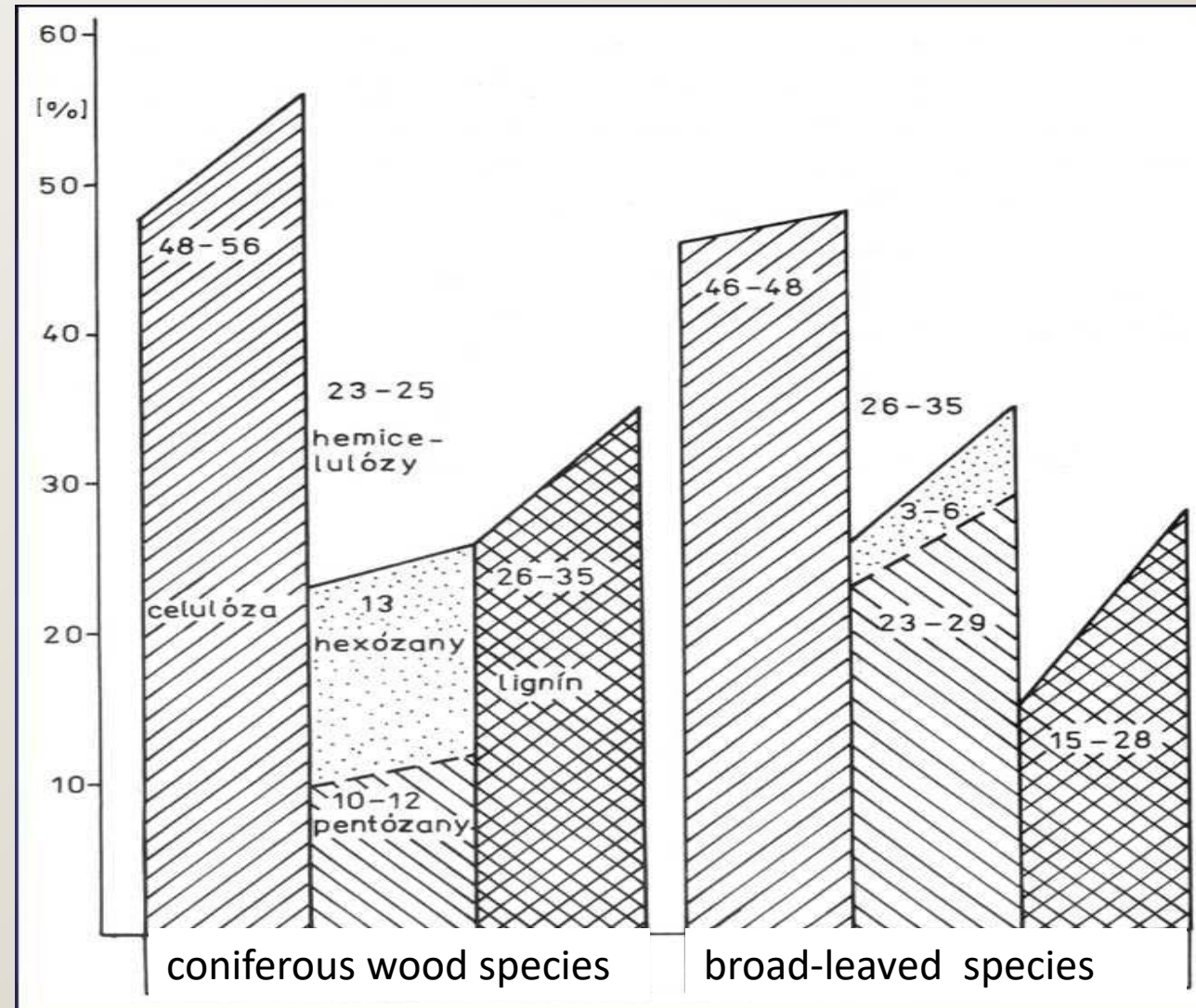
Methoxyl content is used to characterize lignins. Methoxyl analysis of spruce (*Picea abies* (L.)) milled wood lignin indicates a composition  $C_9H_{7.92}O_{2.40}(OCH_3)_{0.92}$ . Beech (*Fagus silvatica* L.) milled wood lignin has a composition  $C_9H_{7.49}O_{2.53}(OCH_3)_{1.39}$

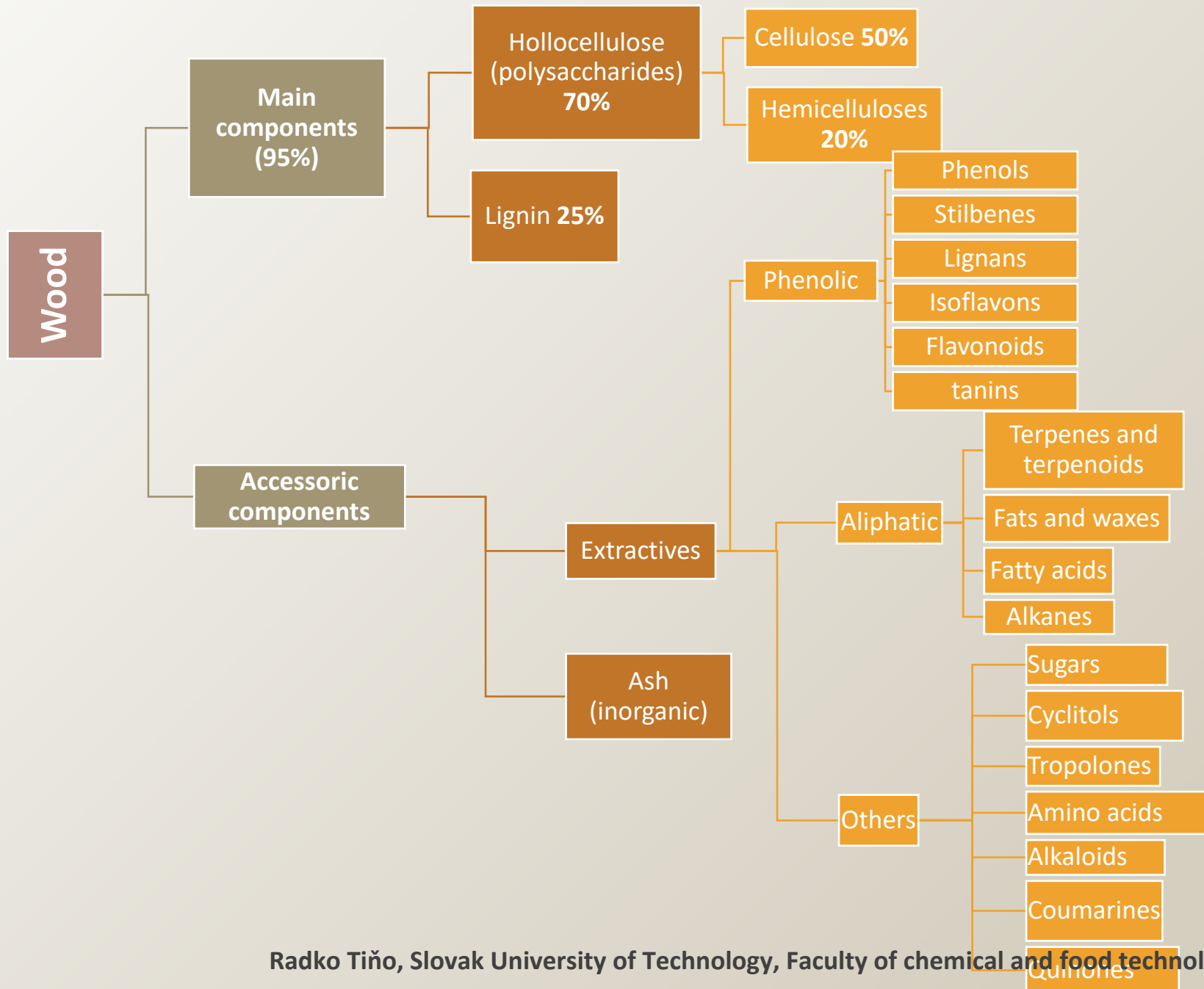
The tubular cell wall has a layered structure:



The **open space in dry wood** is approximately **50%**, but can be as high as **92% in balsa wood**.

# Comparison of main compounds in wood

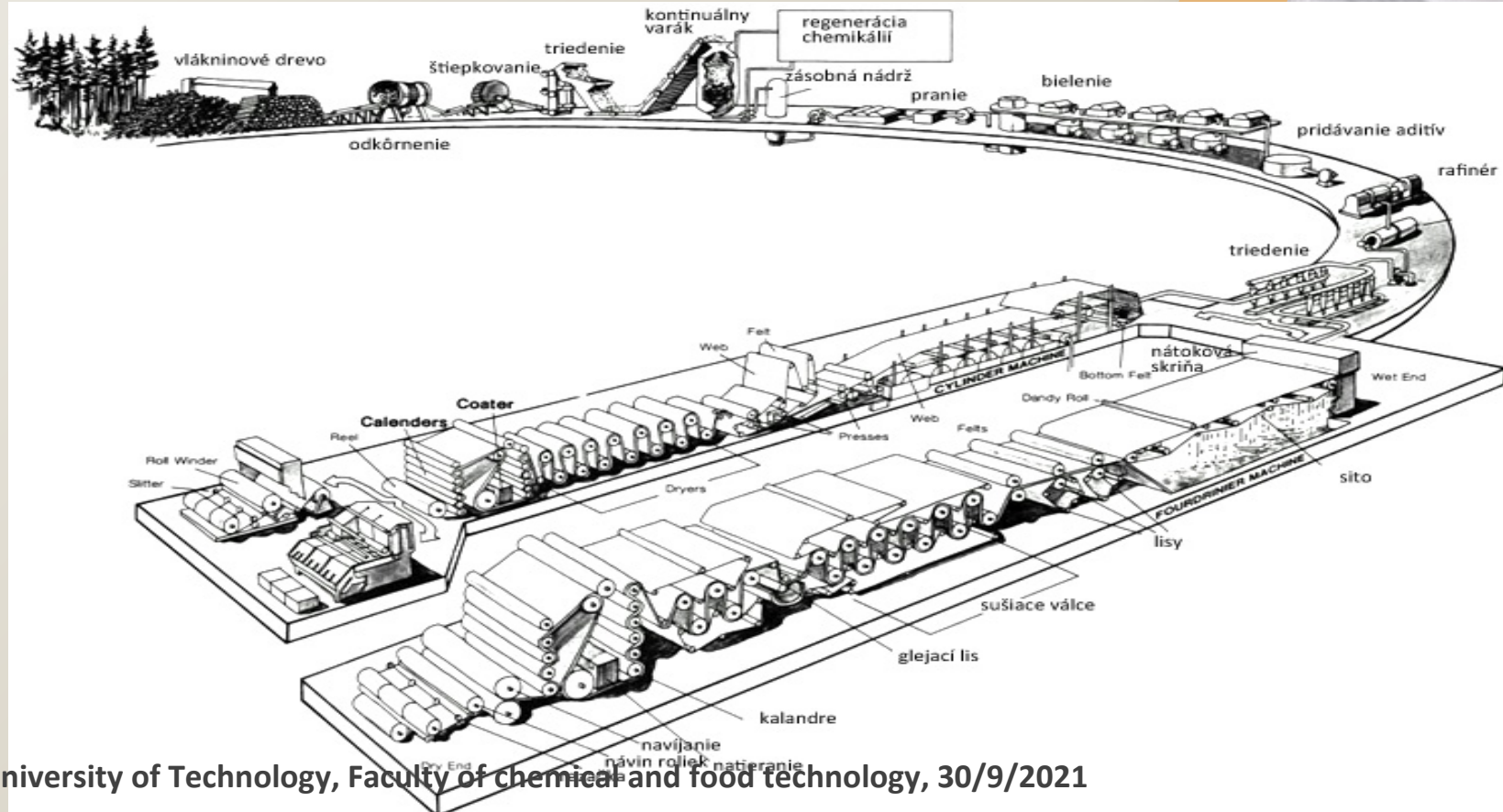
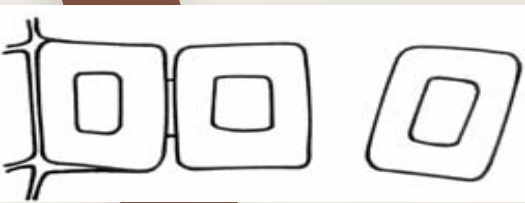
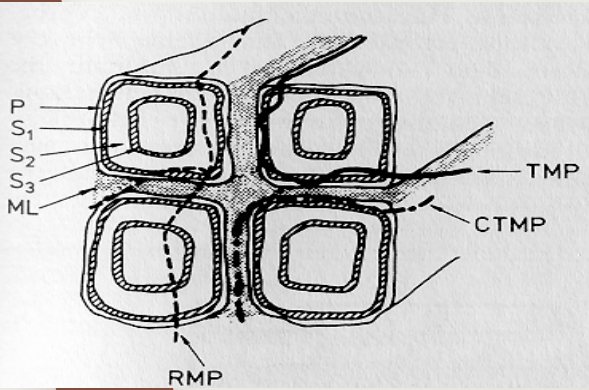
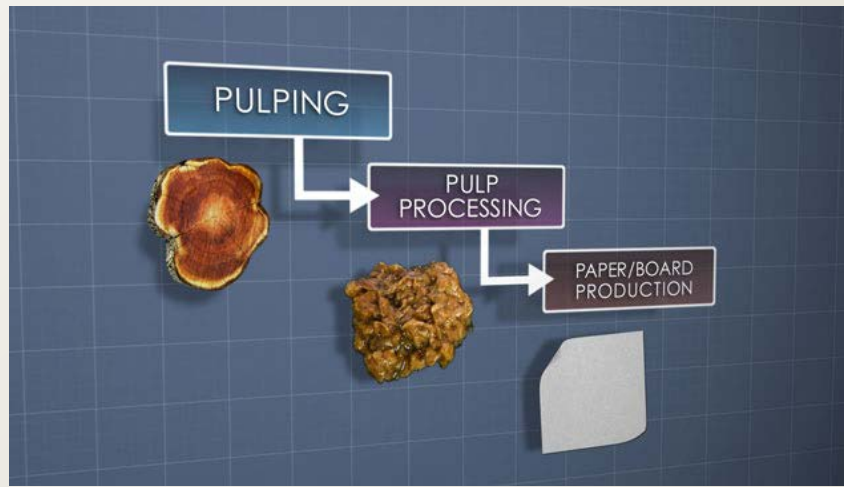


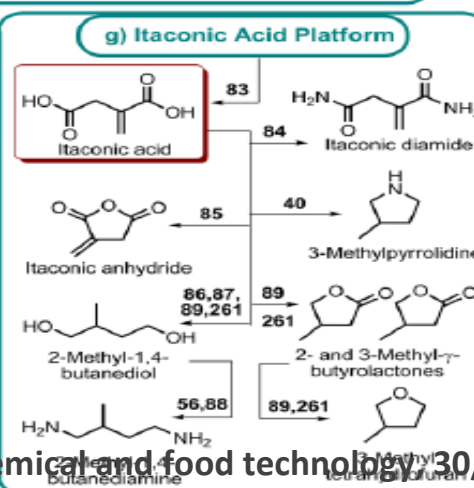
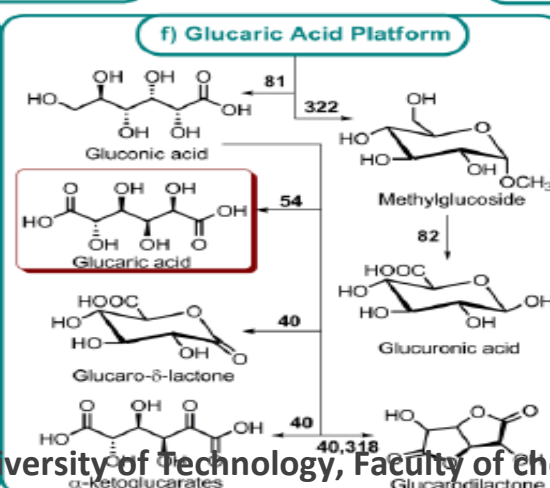
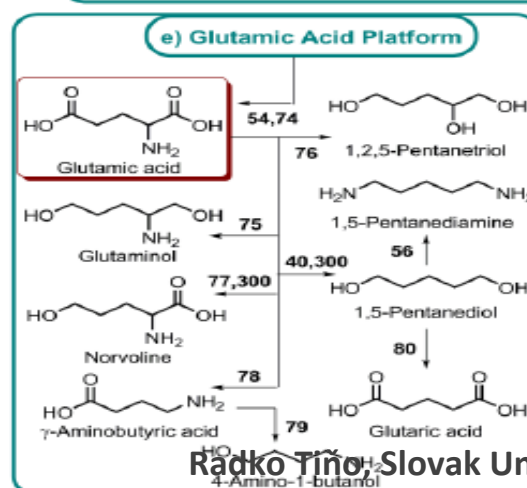
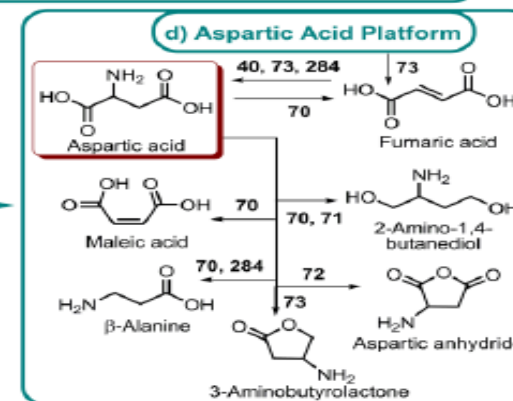
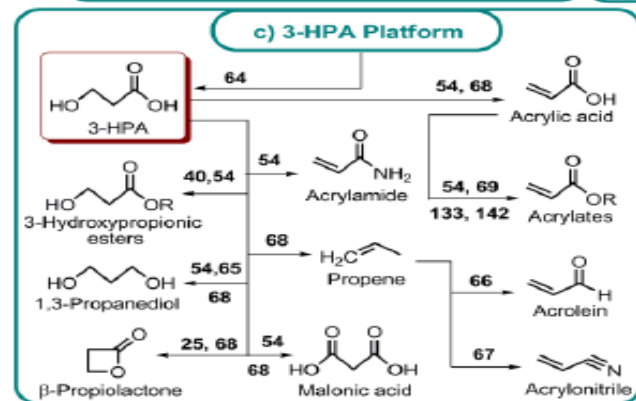
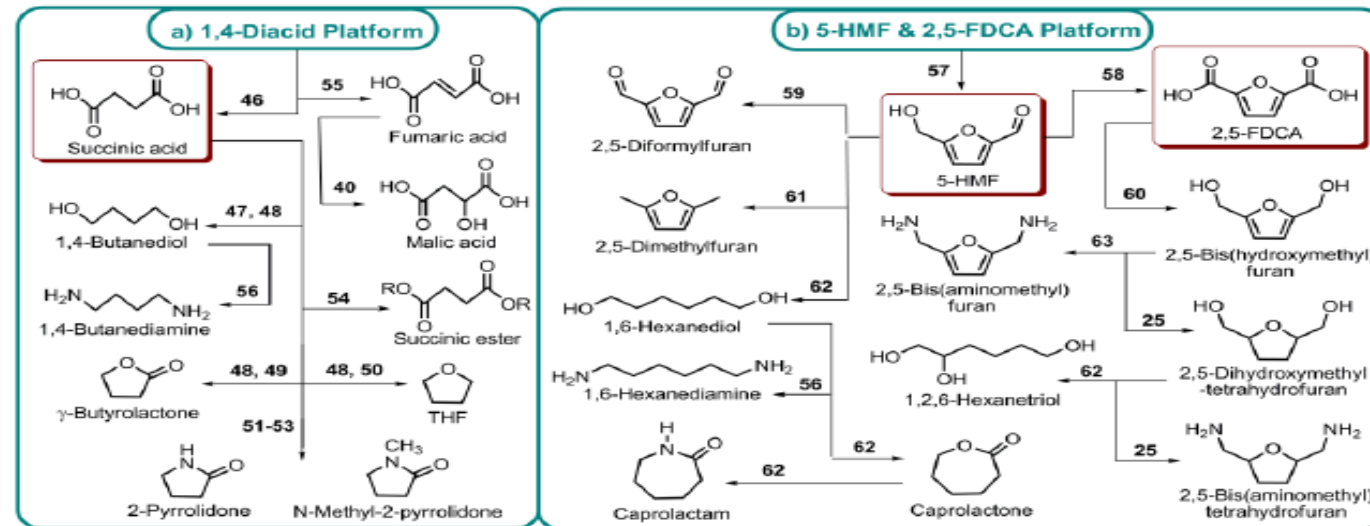


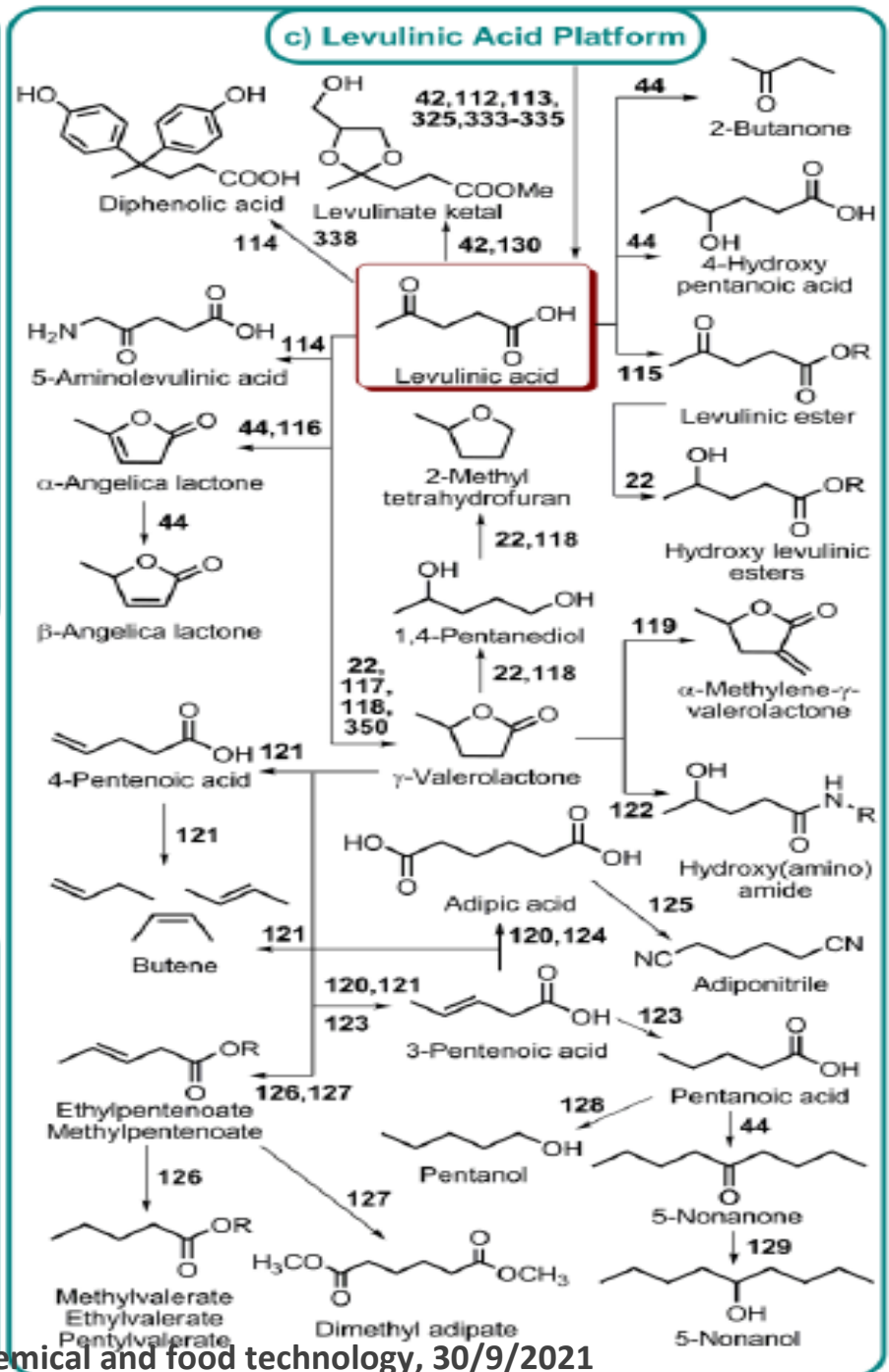
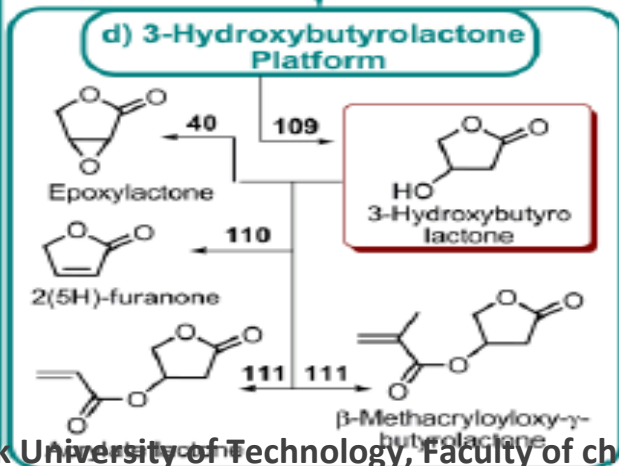
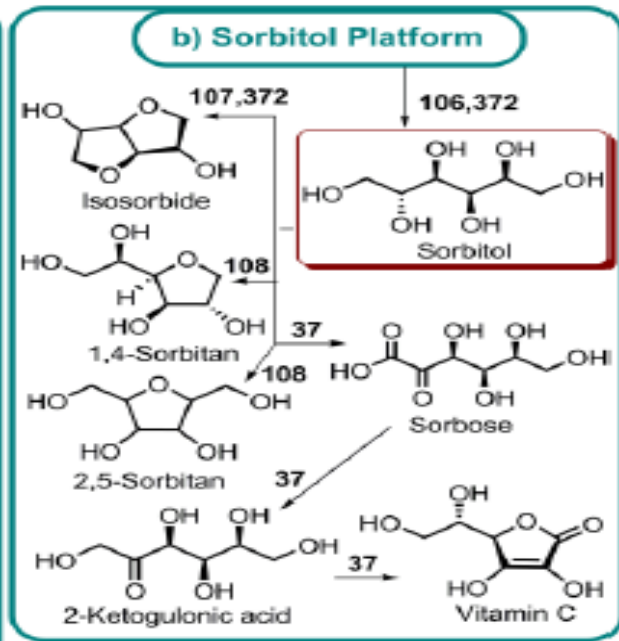
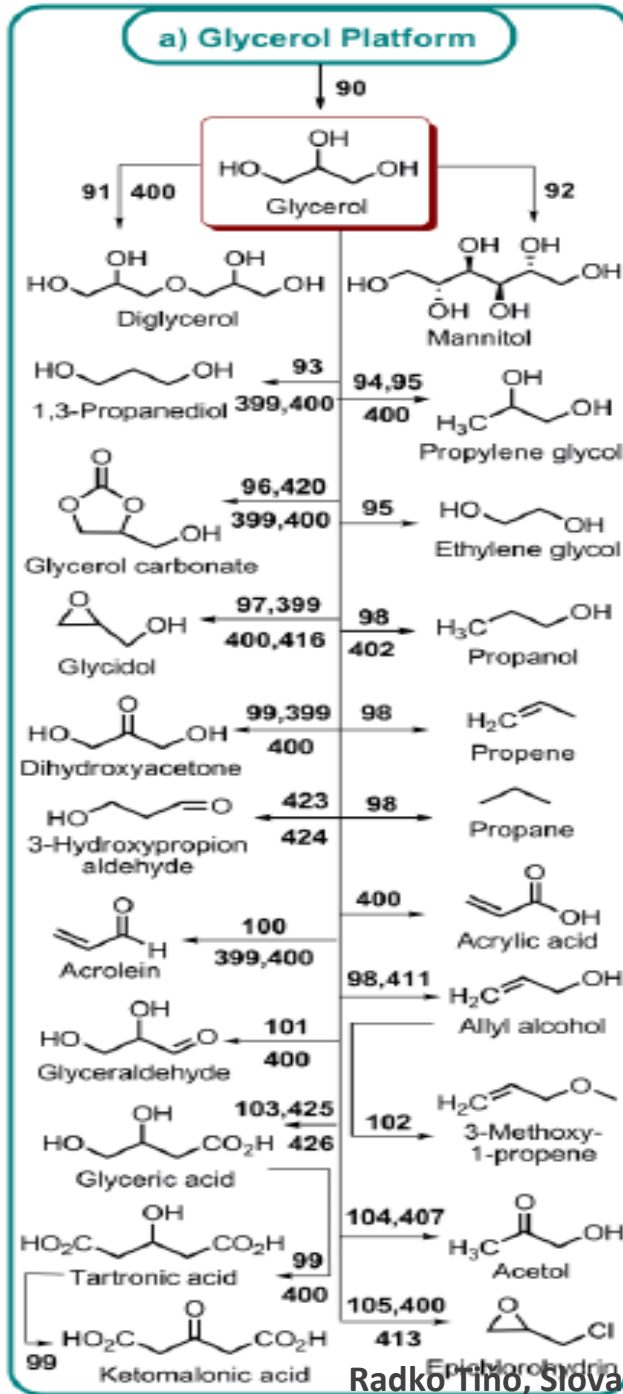
# Chemical composition of main wood species in Central Europe

Component	Spruce european ( <i>Picea excelsa</i> )	Pine common /Scots pine ( <i>Pinus silvestris</i> )	Beech copper ( <i>Fagus silvatica</i> )	Common Oak ( <i>Quercus robur</i> )
ash	0,8	0,4	1,2	0,6
lignin	26,9	26,6	20,9	22,6
$\alpha$ -cellulose	45,6	43,2	39,2	46,3
hemicelluloses	27,6	28,0	35,3	36,7
- manose	16,0	12,5	1,5	-
- xylose	9,0	13,0	28,0	-
extractives	1-4,5	3-5	4,7	2,3-5,6

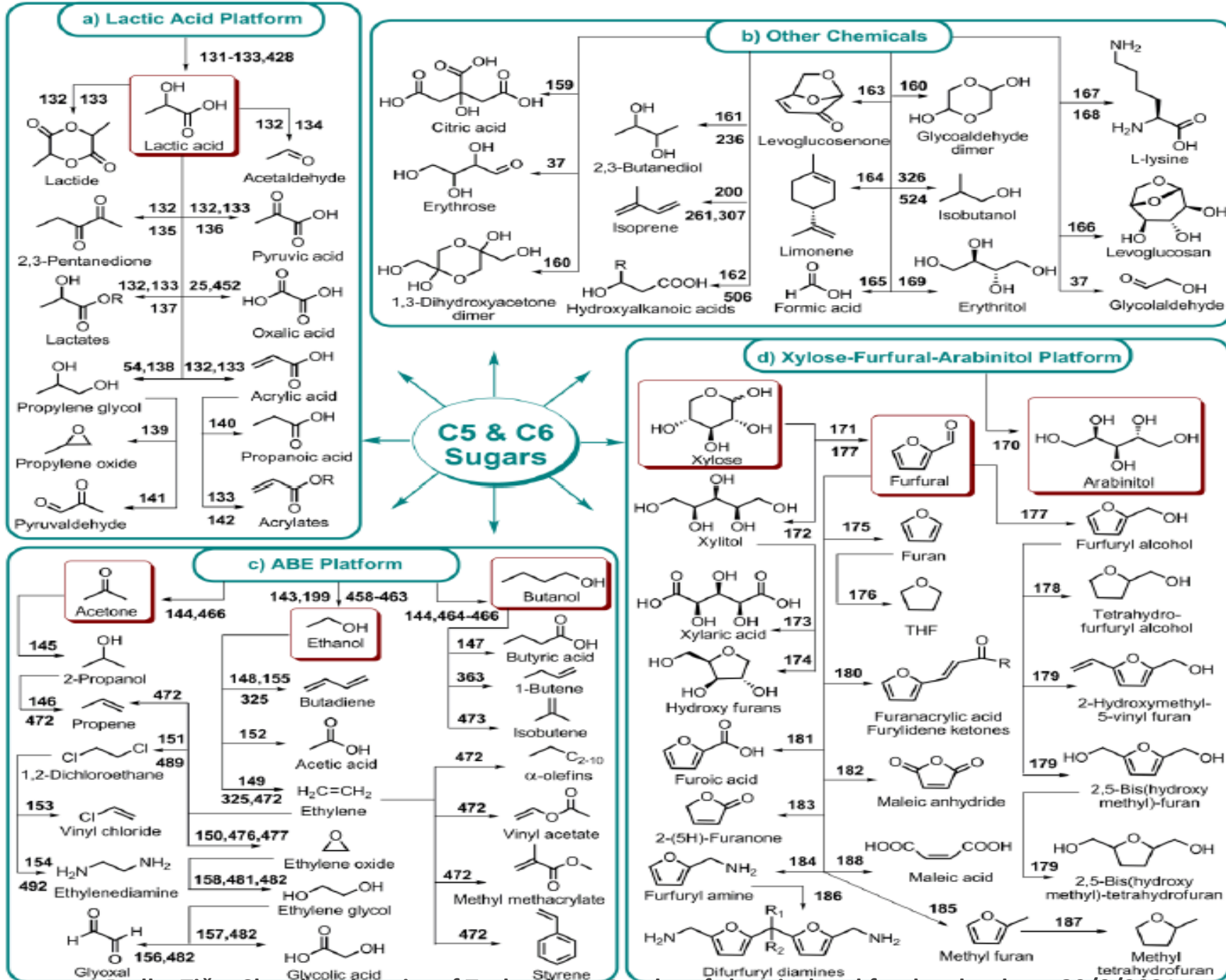
# Pulping and papermaking

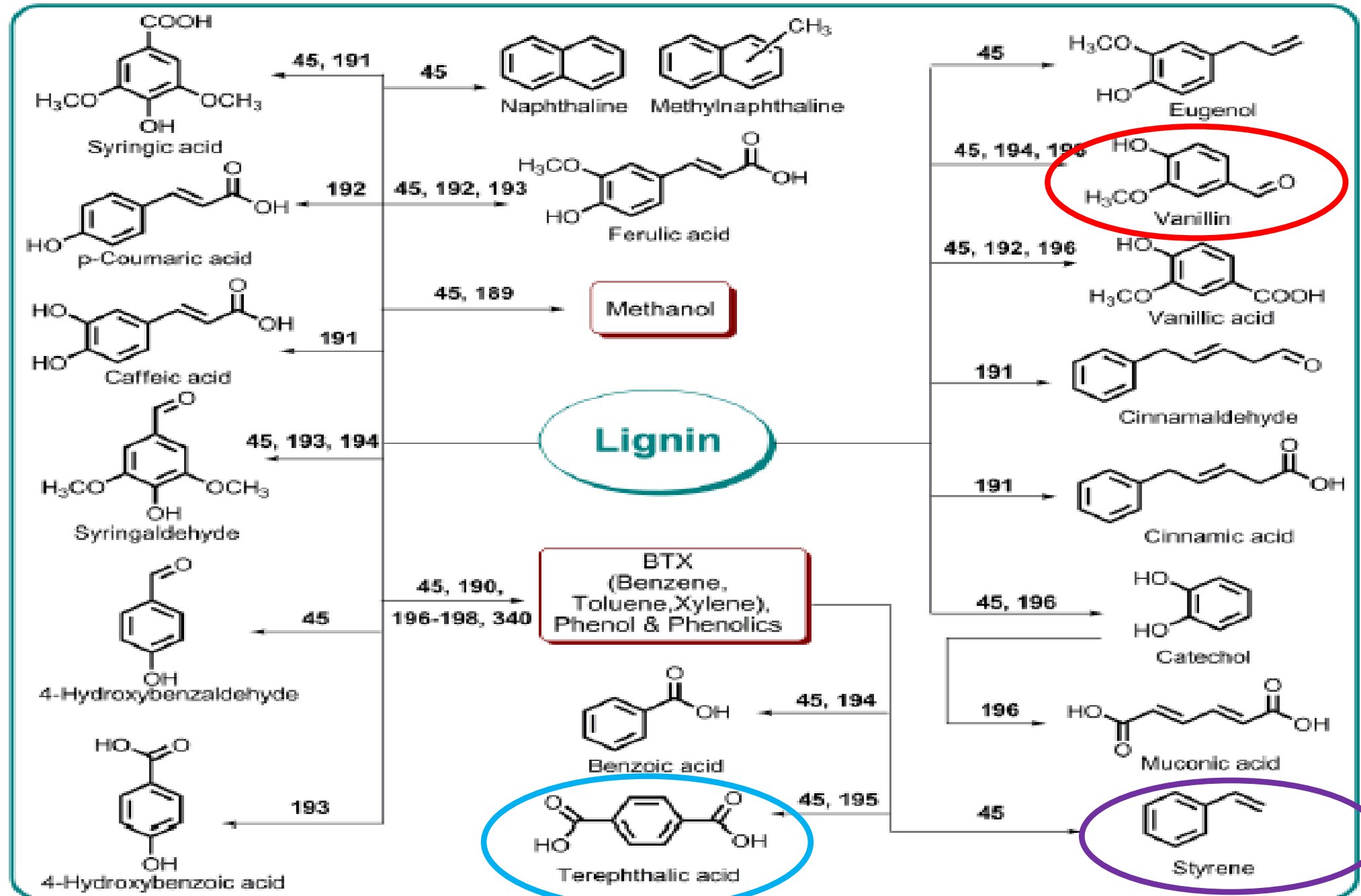












# Wood and water

Water is a **natural constituent** of all parts of a living tree.

In the xylem portion, water (moisture) commonly makes up over half the total weight.

Properties of wood depends on the water content of the wood:

- dimensions
- strength
- thermal conductivity
- resistance to fungi and molds
- electrical resistance
- dielectric properties

# Wood and water

The mass of water in a **freshly felled tree** is **60 to 200%** of the dry mass of the tree.

In **dried out timber** there is only roughly **10 weight percent water content**.

However timbers tend to **achieve equilibrium with the surrounding air**, settling to a moisture content of **22 to 23%** in moist, water-saturated air.

# Location of water in wood

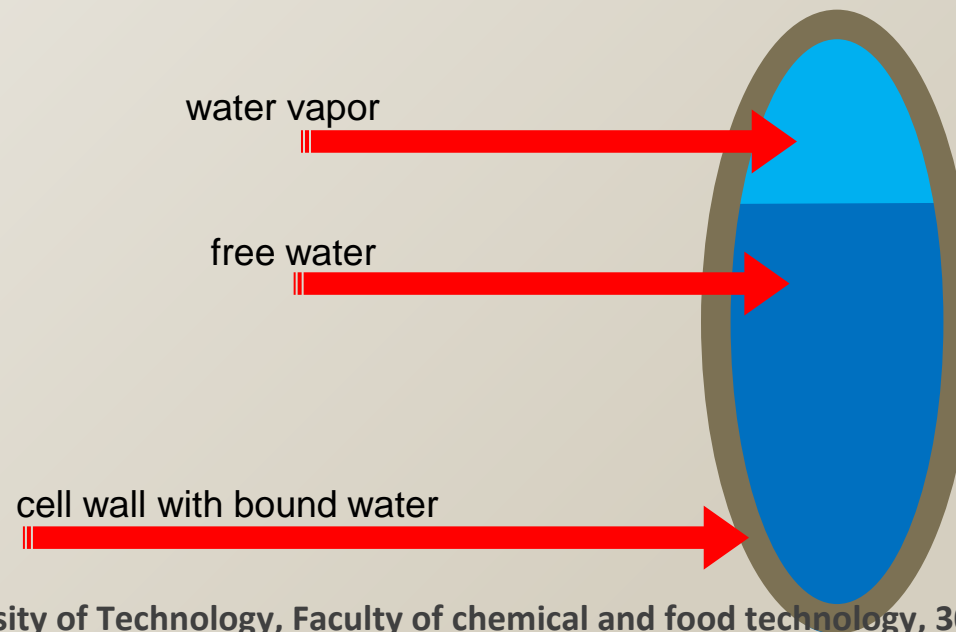
## Location

### In the cell walls – **bound water**

Water can stay in the cell walls at any moisture content level.

### In the cell lumens – **free water**

Water stays in the cell lumens only when the wood is saturated, in other words, when the moisture content of wood is above the fiber saturation point (FSP).



# Nature of water in wood

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## Free water

stayed in the cell lumens  
held by the force of surface tension  
easy to remove, just like the water in a pool

## Bound water

stayed in the cell walls  
held by adsorption force through H-bond  
difficult to remove  
H-bond force increases as MC decreases

## Conception of absorption and adsorption

Absorption: results from surface tension forces.

Adsorption: involves the attraction of water molecules to hydrogen bonding sites.

## Wood is hygroscopic.

- This means the **attraction between dry wood and water** is so **strong** it is impossible to prevent moisture gain.
- Water easily binds with the cellulose fibers (microfibrils) in the cell wall
- As wet wood dries, **free water leaves the lumens (cell cavities) first**. *Free water resembles liquid in a bucket. When you dump water out of a bucket, the bucket does not change shape. Similarly, wood does not shrink as it loses free water from the lumen.*
- After all the free water is gone and only bound water remains, the cell has reached its fiber saturation point (fsp).

## Fiber saturation point

### Definition

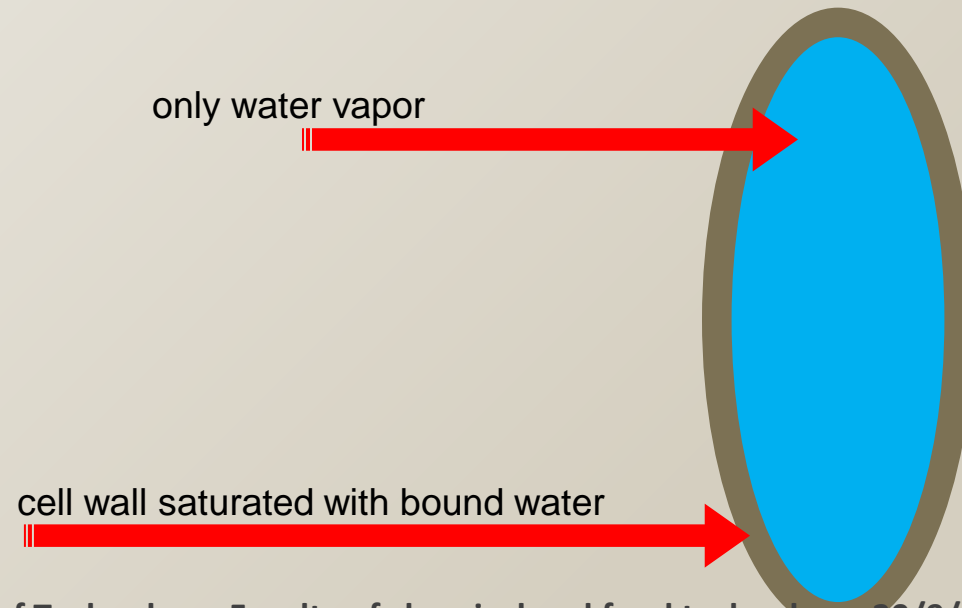
The point at which all the liquid water in the lumen has been removed but the cell wall is still saturated is termed the **fiber saturation point (FSP)**.

### Implication

This is a critical point, since below this point almost all properties of wood are altered by changes in moisture content.

### Ranges

MC = 28-35%, usually take it as 30%.

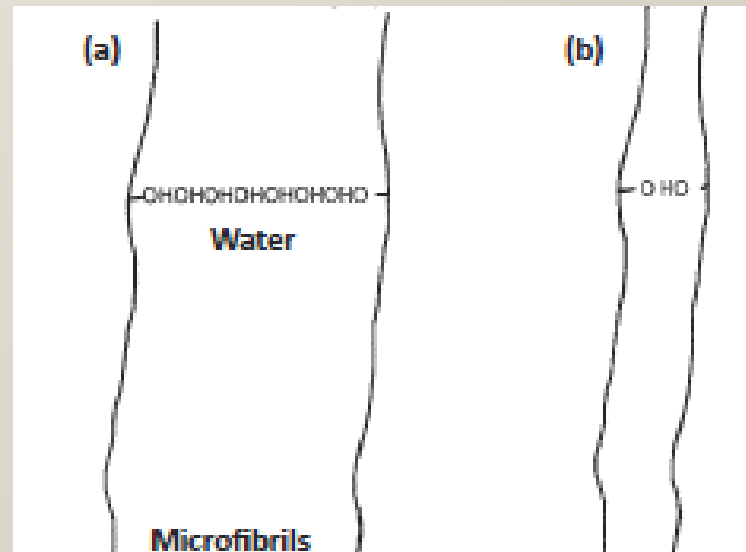




At this point, no water is present in the cell lumen, but the cell wall is completely saturated. It can hold no more water between the microfibrils.

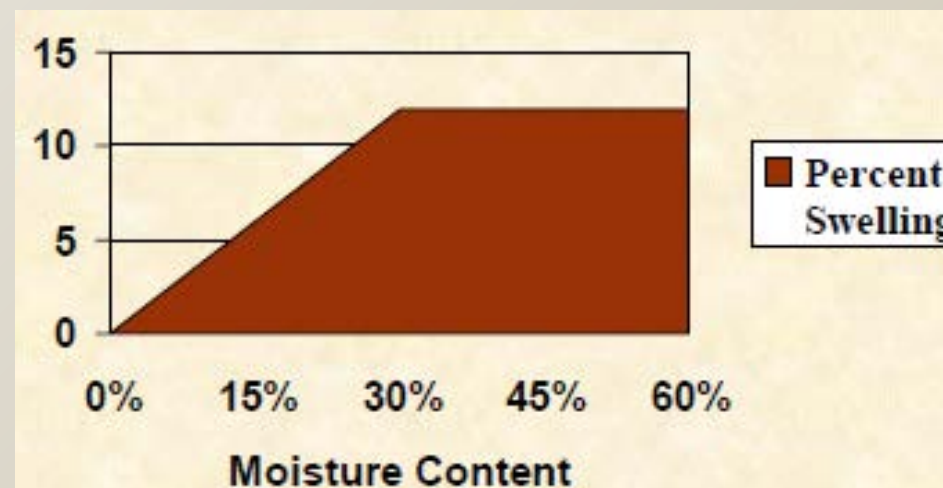
As wood is dried further, bound water leaves the cell wall, and cells start to lose moisture below the fsp.

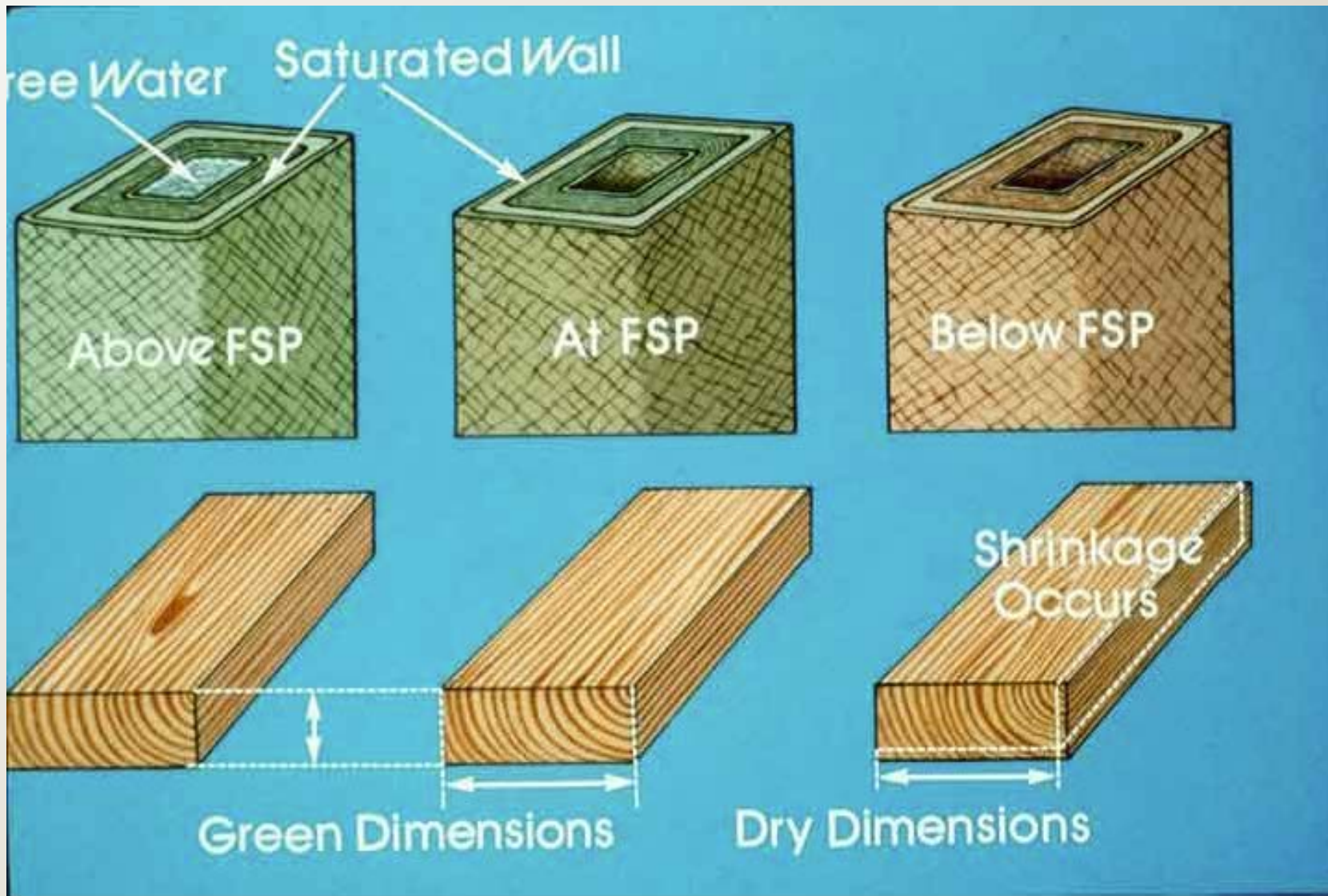
As water leaves and the microfibrils come closer together, **shrinking occurs**.



When **moisture is added to wood**, the process is **reversed**.  
First, water enters the spaces between the microfibrils in the cell wall. The wood swells as the microfibrils are forced apart.

Once the fsp is reached, excess moisture **re-enters the wood lumens** and is held like water in a bucket. This water does not force the microfibrils apart since they are as far apart as possible at the fsp. Wood does **become heavier** as moisture is added above the fsp **but it does not swell** any further.



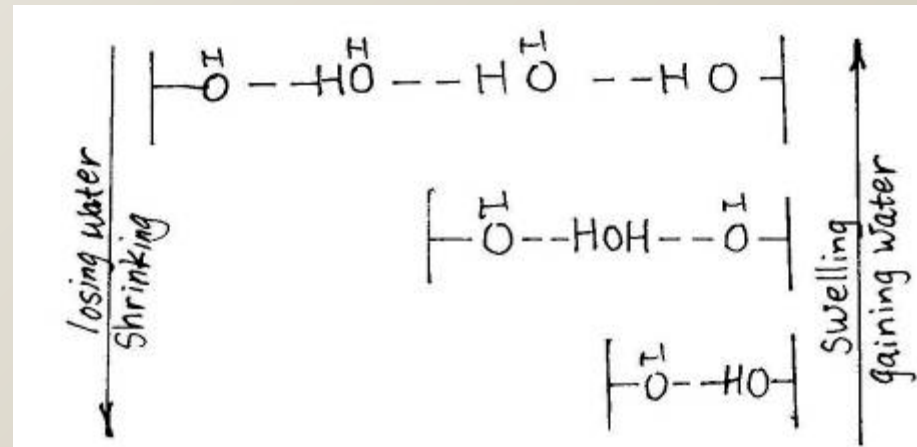


## Not completely reversible in swelling and shrinkage

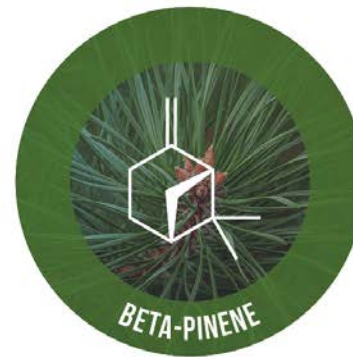
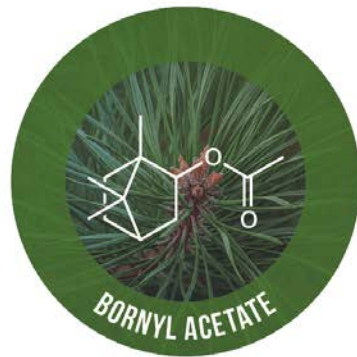
- due to the two different reference bases
- due to the hysteresis effect in the adsorption process

## Mechanisms of shrinking and swelling of wood

- **Shrinking**: the collective effect of cellulose chains moving closer when losing water.
- **Swelling**: the collective effect of cellulose chains moving apart when gaining water.



# Aroma of the Christmas tree



## BORNYL ACETATE

FORMULA  
 $C_{12}H_{20}O_2$

CLASS OF COMPOUND  
Ester

**ABOUT**  
Found in volatile oils of conifers, particularly silver pines and balsam firs, and has a clean, pine-like odour. The use of silver pines as Christmas trees has declined, but balsam firs are popular in the United States.

## ALPHA-PINENE

FORMULA  
 $C_{10}H_{16}$

CLASS OF COMPOUND  
Terpene

**ABOUT**  
Alpha-pinene has a turpentine-like odour, and is found in volatile oils of many different species of pine tree. It has two enantiomers; one is more common in European pines, the other more common in North American pines.

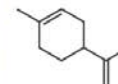
## BETA-PINENE

FORMULA  
 $C_{10}H_{16}$

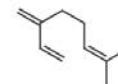
CLASS OF COMPOUND  
Terpene

**ABOUT**  
An isomer of alpha-pinene, beta-pinene is a major compound emitted by forest trees. It has a fresh, woody, turpentine-like smell. Both pinene isomers are flammable, hence why pine cones & Christmas trees burn well.

## OTHER COMPOUNDS



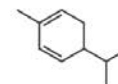
LIMONENE



MYRCENE



CAMPHENE



$\alpha$ -PHELLANDRENE



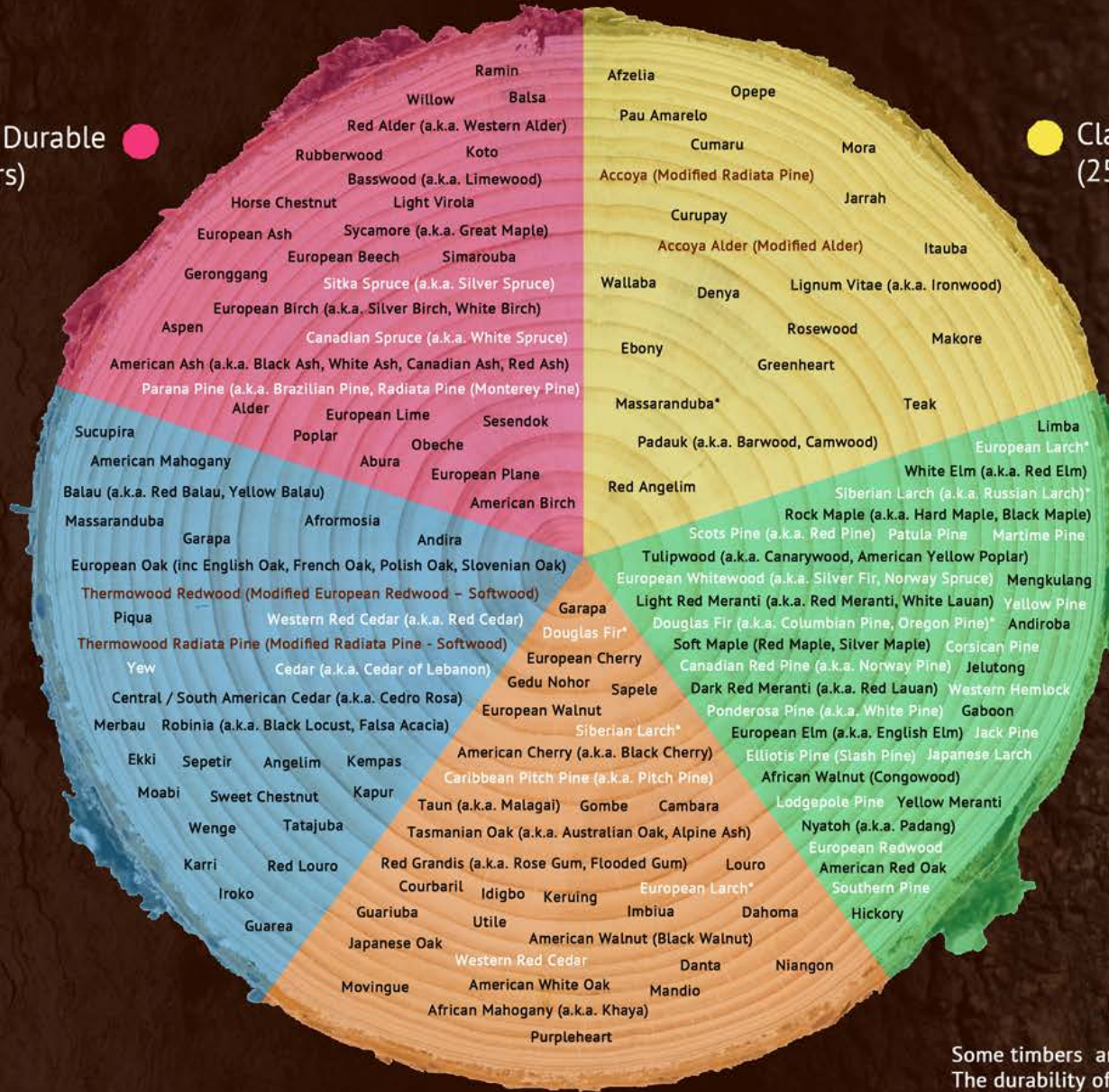
# TIMBER DURABILITY CHART

Class Five - Non Durable  
(Less than 5 years)

Class One - Very Durable  
(25 years+)

Class Two - Durable  
(15 to 25 years)

Class Four - Slightly Durable  
(5 to 10 years)



Some timbers are listed in more than one Durability Class. The durability of these timbers is affected by where they were grown.

GRACIAS  
ARIGATO  
SHUKURIA  
JUSPAXAR  
DANKSCHEEN  
TASHAKKUR ATU  
YAQHANYELAY  
SUKSAMA  
EKHMET  
GRAZIE  
MEHRBANI  
PALDIES  
YOU  
BOLZIN  
MERCII  
BIYAN  
SHUKRIA  
TINGKI



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